Fast, microfabricated, normal phase TLC plates based on carbon nanotube forest scaffolds

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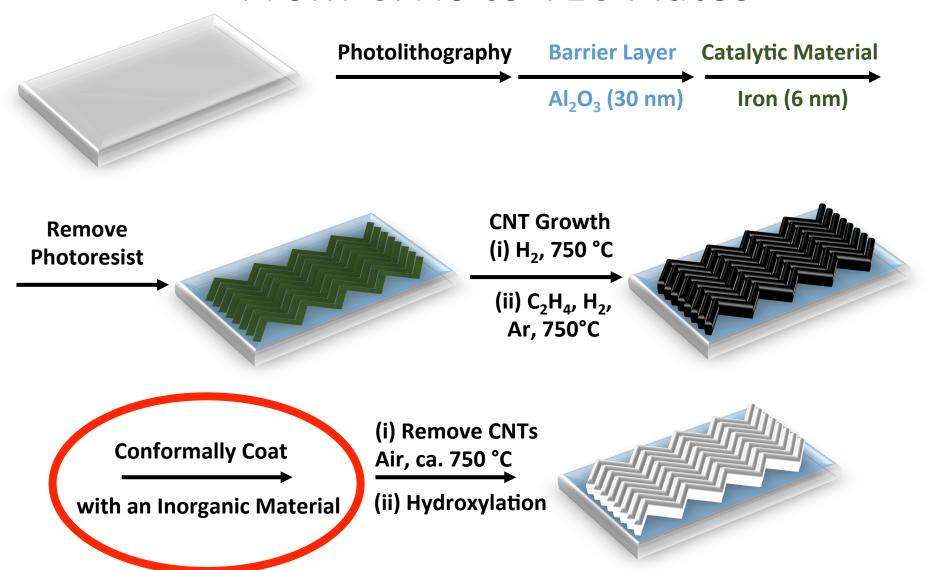
1. Brigham Young University, 2. Diamond Analytics, 3. Justus Liebig University Giessen

Latest Advances in Preparing Microfabricated TLC Plates from Patterned CNT Scaffolds

Three new advances

- LPCVD of silicon nitride
 - Fast
 - Very robust plates
 - SiO₂
- Glass substrates
- Fluorescent substrates

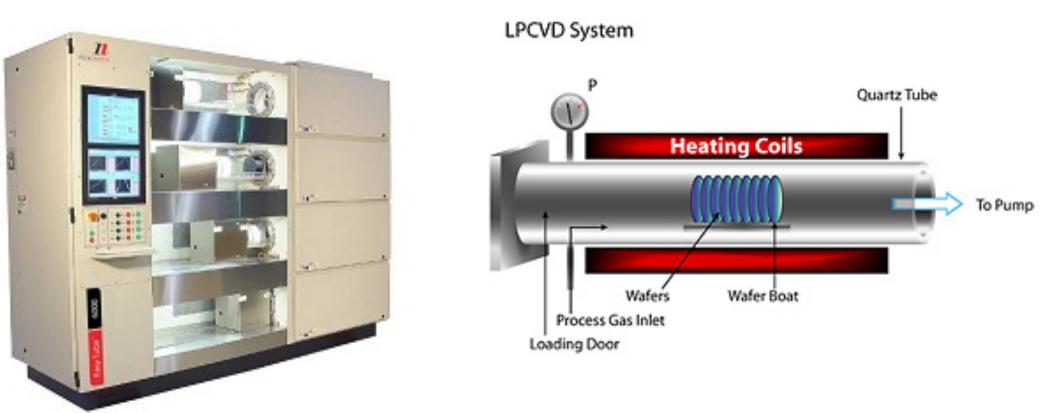
From CNTs to TLC Plates



D. S.; Kanyal, S. K.; Gupta, V.; Vail, M. A.; Dadson, E. A.; Engelhard, M.; Vanfleet, R.; Davis, R. C.; Linford, M. R. Stable, microfabricated thin layer chromatography pl distortion on patterned carbon and Al₂O₃-primed carbon nanotube forests. *J. Chromat. A.* **2012**, 1257, 195-203.

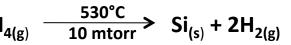
Low Pressure Chemical Vapor Deposition

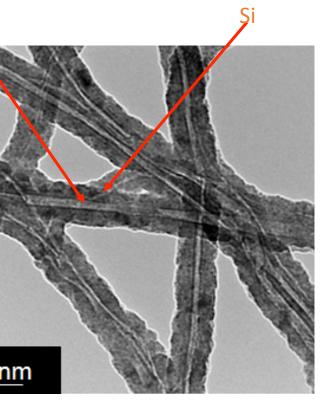
Reagent gasses continuously flow, and volatile biproducts are continuously remove



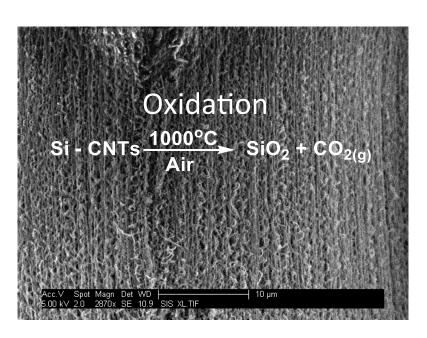
 $Images \ from \ 1. \ http://www.dowcorning.com/content/etronics/etronicschem/etronics_newcvd_tutorial3.asp?DCWS=Electronics&DCWSS=Chemical+Vapor+Deposition \ 2. \ http://sindhu.ece.iisc.ernet.in/nanofab/twikii/bin/view/Main/LPCVD$

LPCVD of *Silicon* onto CNTs



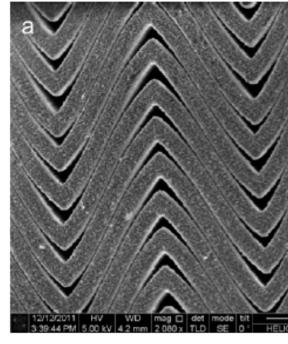


TEM image of silicon coated CNTS



Song, J.; Jensen, D. S.; Hutchison, D. N.; Turner, B.; Wood, T.; Dadson, A.; Vail, M. A; Linford, M. R.; Vanfleet, R. R.; Davis, R. C.; Carbon-nanotube-templated Microfabrication of porous siliconcarbon materials with application to chemical separations. *Adv. Funct. Mater.* **2011**, *21*, 1132-1139

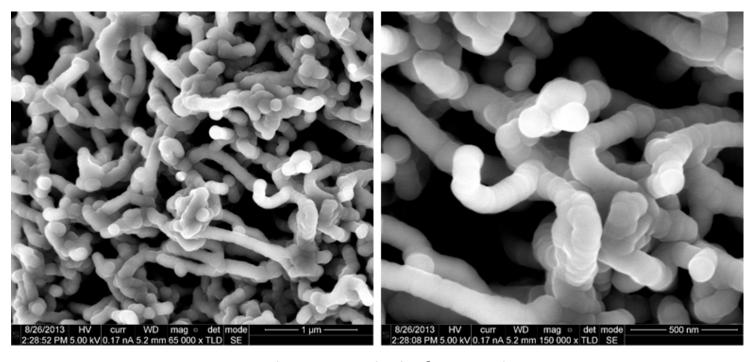
Volume expansion re in feature distorti



SEM of features after oxidation

LPCVD of *Silicon Nitride* onto CNTs

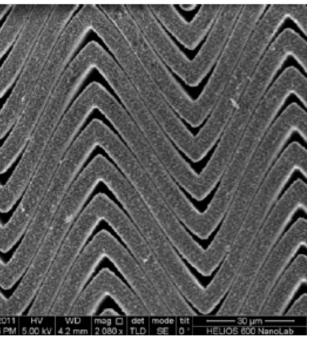
$$SiH_2Cl_{2(g)} + NH_{3(g)} \xrightarrow{780^{\circ}C} Si_3N_{4(s)} + HCl_{(g)}$$
 $Si_3N_{4(s)} \xrightarrow{1000^{\circ}C} Air, 48 h$ $SiO_{2(s)}$



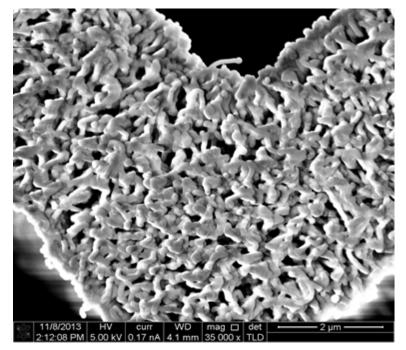
LPCVD silicon nitride before oxidation

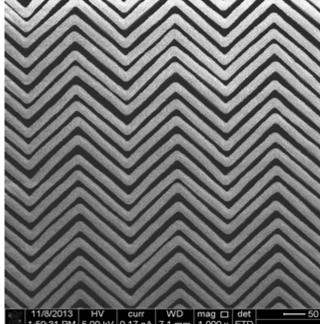
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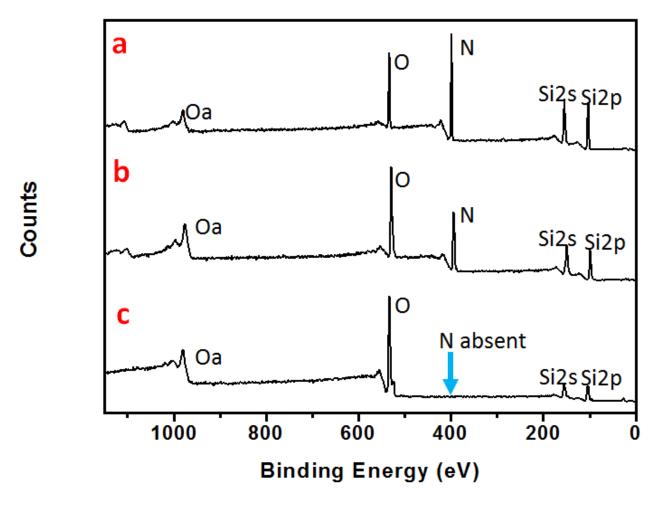




LPCVD silicon nitride after oxidation

Nitrogen Can Be Completely Removed from the Near Surface Region of the Si₃N₄ plates

(a) as deposited Si₃N₄ film (b) oxidized at 600 °C, 48 h c) oxidized at 1000 °C, 48 h



Separation on an LPCVD Silicon Nitride TLC Plate

s: Food dye mixture

tance: 45 mm time: ca. 5 min

ness: Plates can be washed

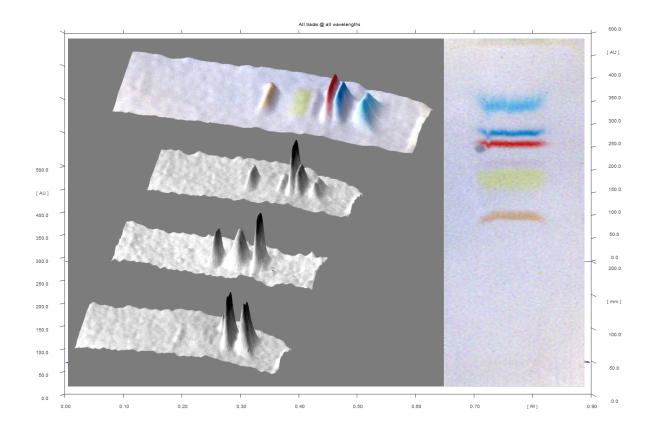
sed dozens of times

ocessing: Color enhanced for

ation

nt: 2 - 3 times as fast as a

rcial plate



Densitometer scan at different wavelengths

Separations from our Lab at BYU on LPCVD Silicon Nitride TLC Plates

ication: Plates (i) and (ii) made

different masks

ytes: CAMAG test dye mixture distance: 25 mm (i and ii), 35

(iii)

on time: (i) 1 min 10 s, (ii) 1 min

(iii) 3 min 15 s

size: 3 mm

ile phase: t-butyl benzene

nidity: 21% (typical)

perature: 22 C (typical)

ment: 2 – 3 times as fast as

mercial plates







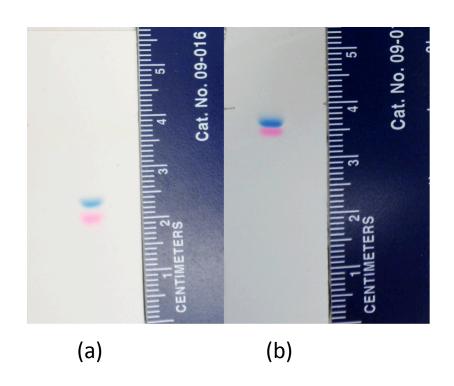
(i)

(ii)

(iii)

Separation of two dyes: BB7 and Rhodamine

- Separation of BB7 and rhodamine dye on
 - (a) Merck TLC plate
 - (b) M-TLC-plates.
 - Development solvent: EtOAc:MeOH:H₂O (75:15:10)
 - Run times: 1 min 15 s and 3 min 43 s for M-TLC and Merck TLC plates, respectively



Two Quick Teasers

Transparent (glass) substrates

Fluorescent plates

Transparent Substrate

eed to find high mperature bstrate Somewhat challenging problem ost glasses soften d/or deform at evated mperatures dapt process to mewhat lower mperatures ansmission mode etection is important r TLC scanners

w Observations:



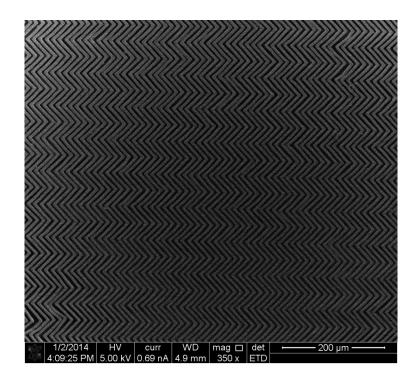
720 °C for 48 h
One substrate unaffected, another bends slightly, another to a significant degree



620 °C for 48 h Flat as a board

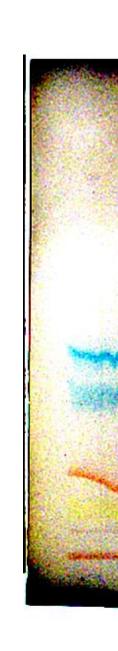
Transparent Substrate

- The lithography changes a little on a different substrate
- Still optimizing



Patterned CNT forests grown on a high temperature glass

- Separation of a food dye mixture on a glass TLC plate.
- Color enhanced for easy visualization of spots.



Making a Fluorescent Plate

Deposited ZnO(s) into the silicon nitride plates

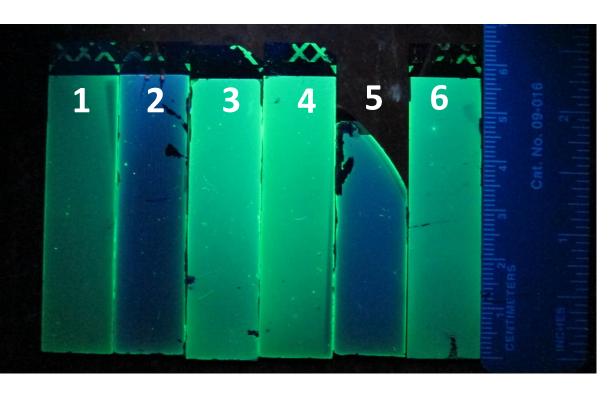
ALD of dimethylzinc (DMZ) and water

Depending on deposition conditions, plates show green fluorescence

Fluorescence good with 254 nm excitation

Analytes on the plates quench the fluorescence

The chromatography doesn't change – same selectivity



Analytes: caffeine (1) and phenacetin (2)

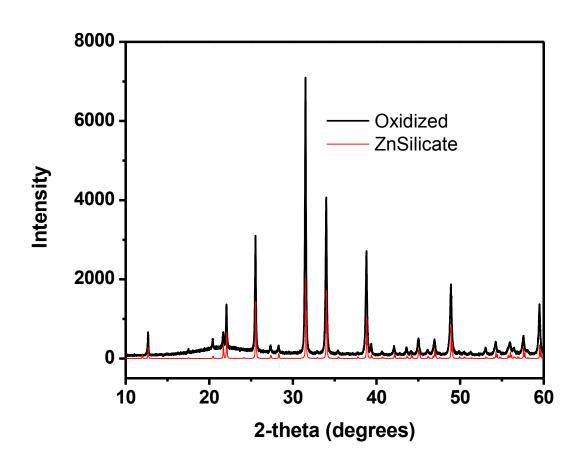
Mobile phase: chloroform:methanol:acetic acid (80:15:5 v/v/v)

1

7

Making a Fluorescent Plate

Intriguing result:



Conclusions

LPCVD of silicon nitride leads to good silica TLC plates

- No deformation of features
- No nitrogen left by XPS after oxidation
- Good separations of food dyes, CAMAG dye mixture, and other dyes
- Plates are very robust can be washed and reused multiple times
- Separations comparable in resolution to those on HPTLC plates, but 2 3 times faster

Glass substrates being developed

- Photolithography is possible
- Preliminary separations performed

Fluorescent plates

- Progress towards the production of a fluorescent plate
- Preliminary separations

Acknowledgements

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ndrew Miles

BYU Physics Dept.

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Dr. Richard Vanfleet

BYU Integrated Microfabrication Facility

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