#### **VALIDATED HIGH-PERFORMANCE THIN-LAYER** Université Nice **CHROMATOGRAPHIC METHODS FOR THE DETERMINATION OF S**oph<mark>ia</mark> Antipolis **FLAVOUR COMPOUNDS IN PLANT EXTRACTS**

Lionel PAILLAT <sup>1,2</sup>, Christine PERICHET <sup>1</sup>, Jean-philippe PIERRAT <sup>1</sup>, Sophie LAVOINE <sup>1</sup>, Jean-Jacques FILIPPI <sup>2</sup>, Uwe MEIERHENRICH <sup>2</sup> and Xavier FERNANDEZ <sup>2</sup>

<sup>1</sup> Charabot S.A., 10 avenue Yves Emmanuel Baudoin, 06130 Grasse, France.

<sup>2</sup> Laboratoire de Chimie des Molécules Bioactives et des Arômes, LCMBA UMR 6001 CNRS, Université de Nice – Sophia-Antipolis Parc Valrose, 06108 Nice Cedex 2, France.

**QUANTITATIVE DETERMINATION OF ALCOHOLS AND ACETATES IN** HAITIAN VETIVER EXTRACTS

## INTRODUCTION

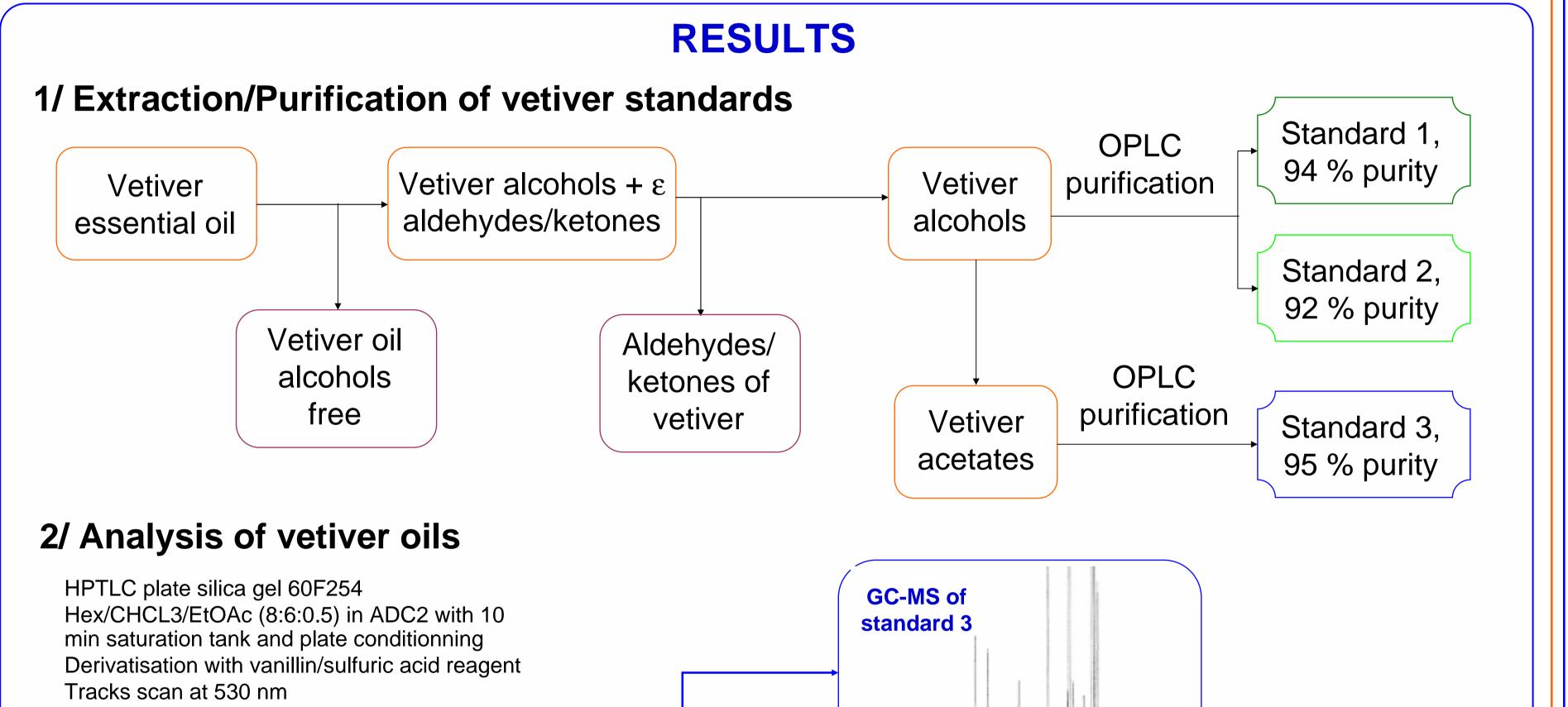
Vetiver grass (Vetiveria zizanioides, Linn Nash) from the family Poaceae is a densely tufted grass growing throughout 70 tropical regions of the world, Java and Haïti being the largest suppliers [1]. Its roots, from which essential oils were obtained, diffuse a flavour that is used in perfumery and flavour industries. Haitian vetiver oil is one of the most complexes with more than 300 sesquiterpenoids [2]. Olfactory studies have shown that alcoholic fraction presents the most interesting flavour of the oil and the corresponding acetylated fraction is more appreciated.

**QUANTITATIVE DETERMINATION OF** VANILLIN  $\beta$ -D-GLUCOSIDE AND FOUR **MAJOR PHENOLIC COMPOUNDS IN VANILLA** FRUITS, BEANS AND EXTRACTS

### INTRODUCTION

Vanilla planifolia is a tropical aromatic orchid widely used in aroma industries for its flavour, mainly due to phenolic compounds [5]. Financial value depends on physical characters of the beans (colour, length and shape). Amount of aroma, especially vanillin, is important in beans quotation. Authenticity of vanilla beans and extracts is checked by determination of phenolic compounds ratio, according to DGCCRF specification [6]; that is why their content have to be determined. Development of vanilla fragrance appears during fruit curing, mainly due to glycosides hydrolysis, like vanillin  $\beta$ -D-glucoside. Its quantification in fruits can help to predict vanillin content in cured beans.

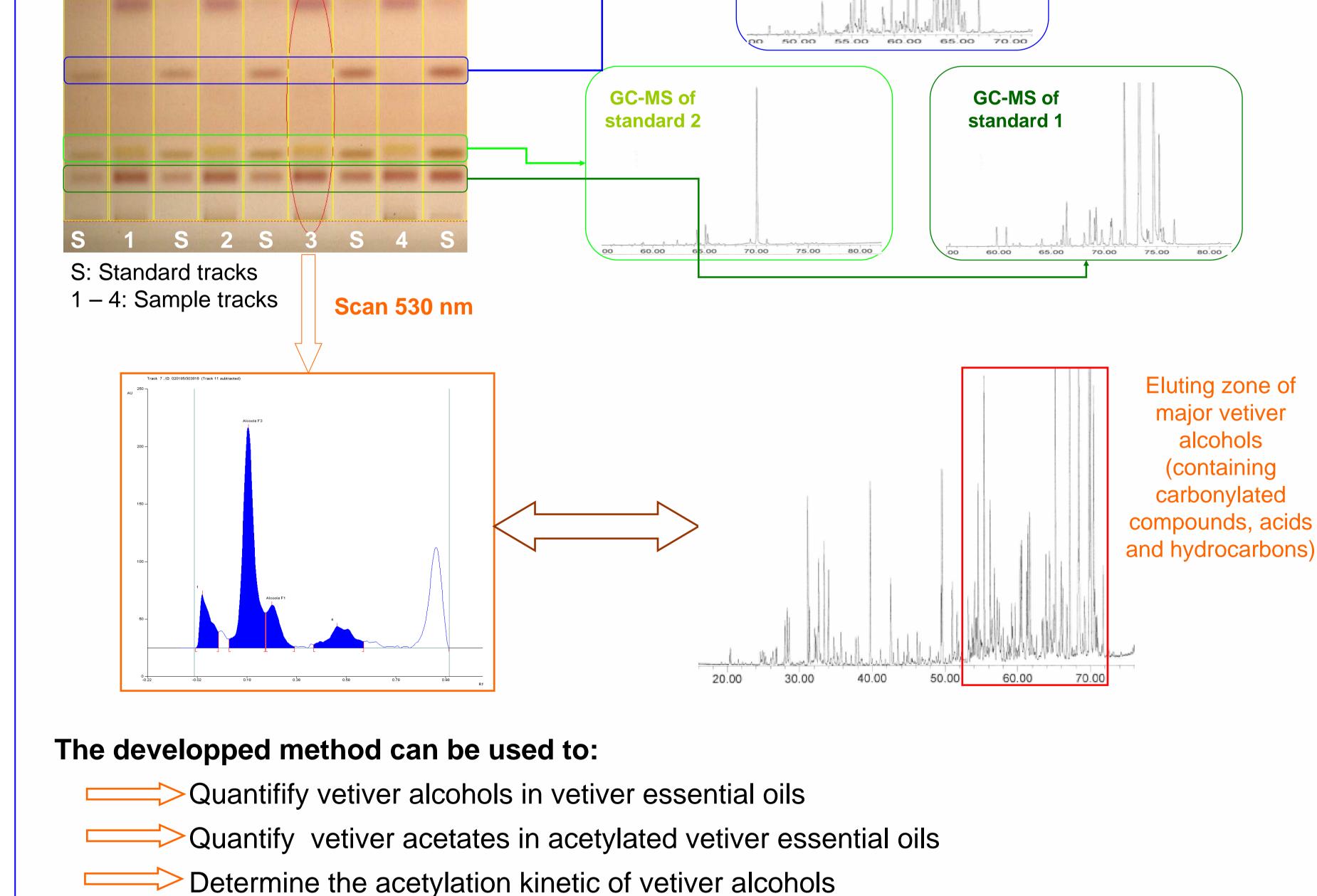
Vetiver alcohols and acetates content are usually determined by GC/FID [3] and GC-MS [4] but precision of these methods are not well established and integration parameters depend on analysts (more than 30 alcohols or acetates). HPTLC analysis can separate compounds on the basis of their functional group (when compounds have the same molar mass); that is why it was decided to use this technique for vetiver alcohols and acetates quantification.

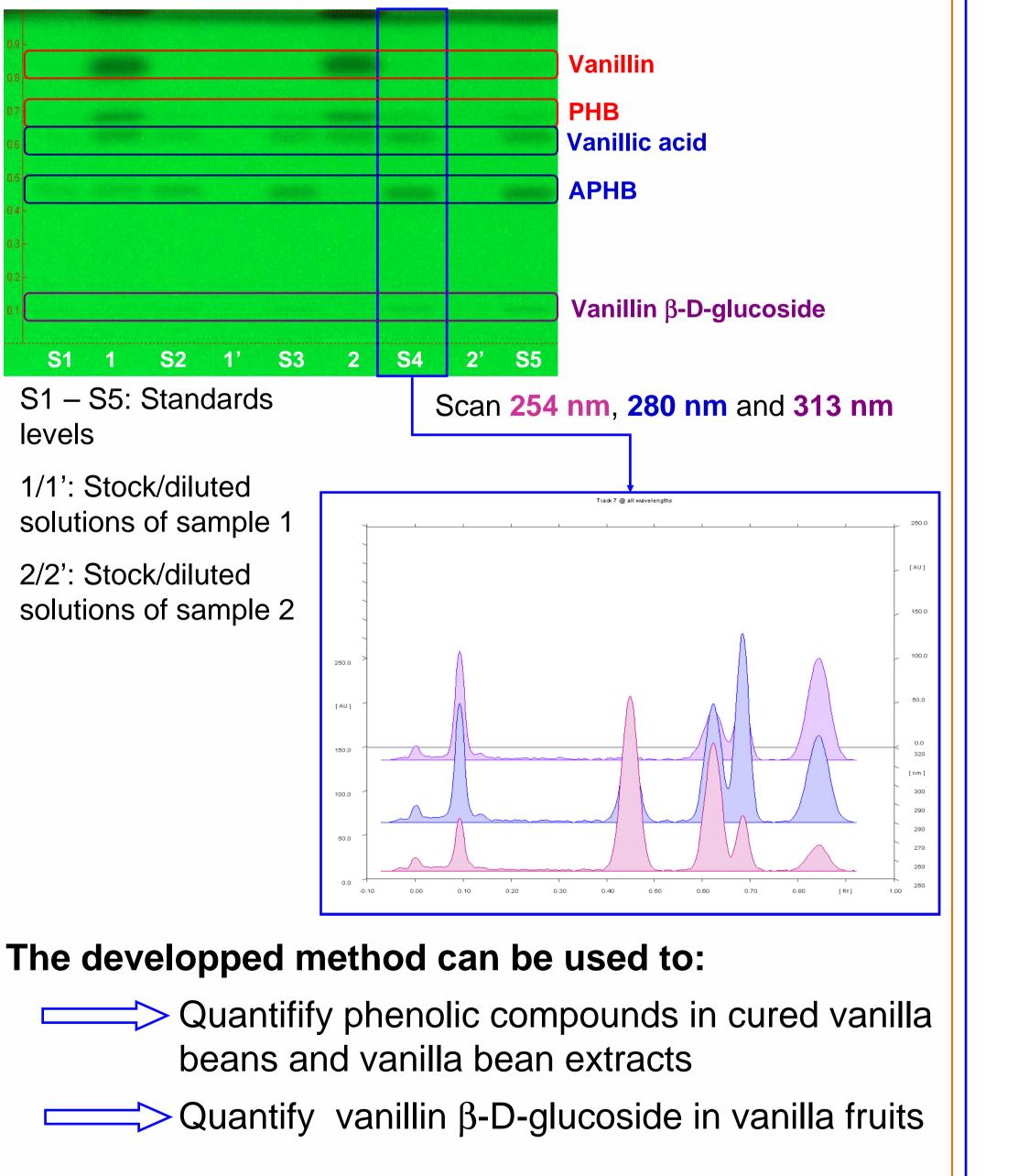


	Expected values	
Vanillin/PHB	10 – 20	РНВ: <i>р</i> -
Vanillin/APHB	40 - 110	hydroxybenzaldehyde
Vanillin/Vanillic acid	12 – 29	APHB: <i>p</i> -
APHB/PHB	0.15 – 0.35	hydroxybenzoic acid
Vanillic acid/PHB	0.55 – 1.5	

RESULTS

HPTLC plate silica gel 60F254 Hex/CHCL3/MeOH/acetic acid (5:36:4:0.5) in ADC2 with 5 min saturation tank and plate conditionning Visualisation at 254 nm Tracks scan at 254 nm (APHB), 280 nm (vanillin  $\beta$ -D-glucoside, PHB, vanillic acid) and 313 nm (vanillin)





#### VALIDATION PARAMETERS

	Compounds	Rf	LOD (ng.spot <sup>-1</sup> )	LOQ (ng.spot <sup>-1</sup> )	Calibration range (ng.spot <sup>-1</sup> )	Recovery (%)
VETIVER	Standard 1	0.18	5	20	40 - 200	98.9
	Standard 2	0.28	10	40	40 - 200	102.0
	Standard 3	0.65	5	30	40 - 200	99.4
VANILLA	Vanillin β-D- glucoside	0.09	8	20	24 – 120	100.4
	APHB	0.42	6	20	21 – 106	98.8
	Vanillic acid	0.57	14	20	20 - 102	98.9
	PHB	0.62	2	6	6.5 – 33	99.5
	Vanillin	0.77	4	8	8 - 40	99.0

# **BIBLIOGRAPHIC REFERENCES**

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Intra- and Inter-day variation of both quantitative HPTLC methods: % R.S.D. < 3 %