

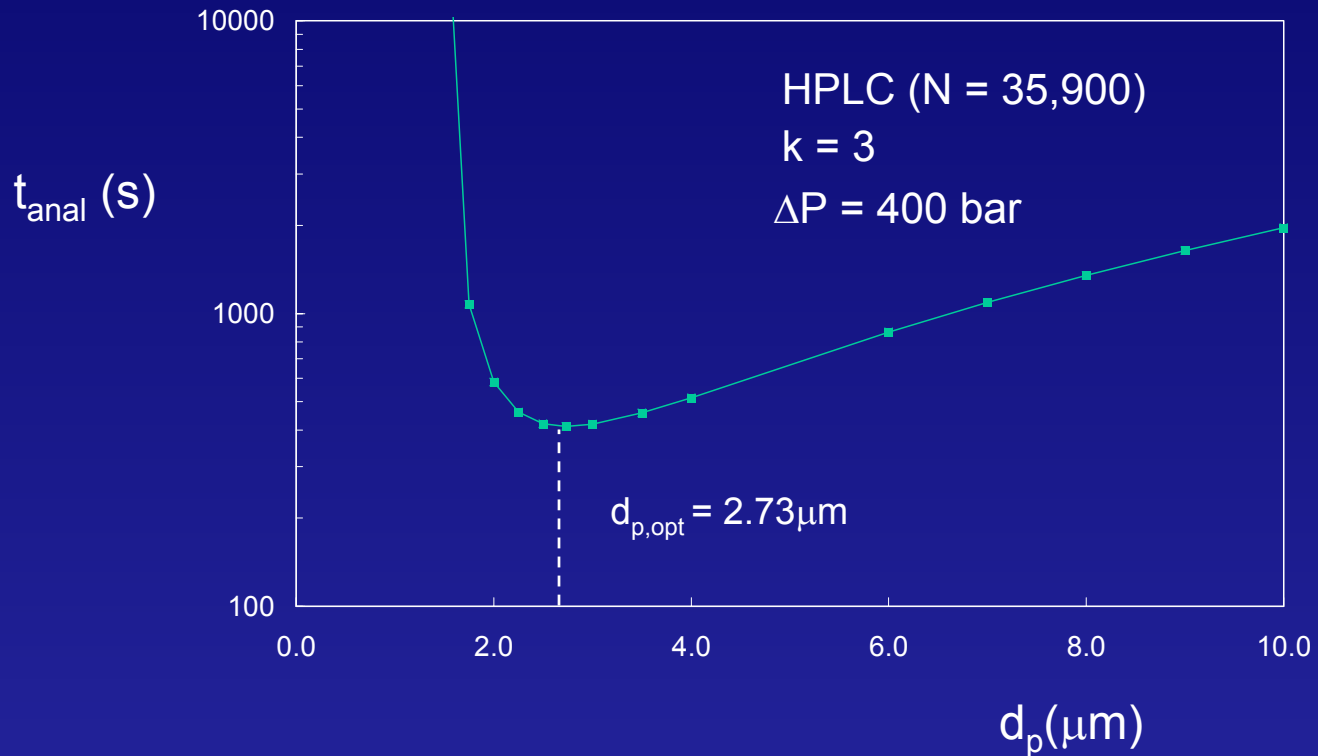
VRIJE UNIVERSITEIT BRUSSEL
Department of Chemical Engineering

Shear Driven Chromatography as a Potential Alternative to HPTLC

Vervoort N., Clicq D., Baron G. and Desmet. G

- 1. Shear-Driven Flows**
- 2. Tracer Flow Experiments**
- 3. Separation Experiments**
- 4. Conclusions**

$$t_{\text{anal}} \sim \frac{N \cdot d_p^2}{D_{\text{mol}}} \quad (\text{when } u = u_{\text{opt}}), \quad \text{but: } u_{\text{opt}} \sim \frac{1}{d_p} \quad \text{and} \quad \Delta P \sim \frac{u \cdot L}{d_p^2}$$

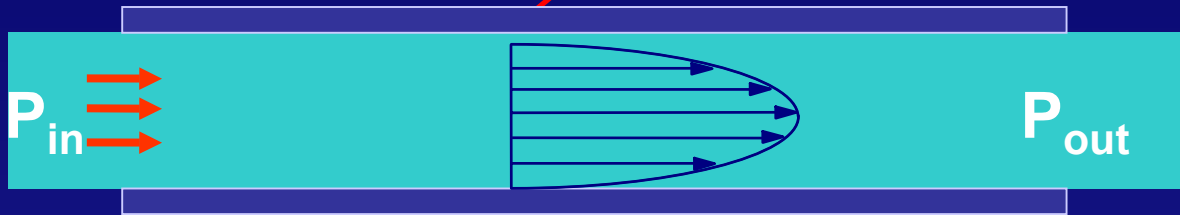


⇒ optimal d-value for minimal t_{anal}

⇒ lower-limit for d

Pressure-Driven:

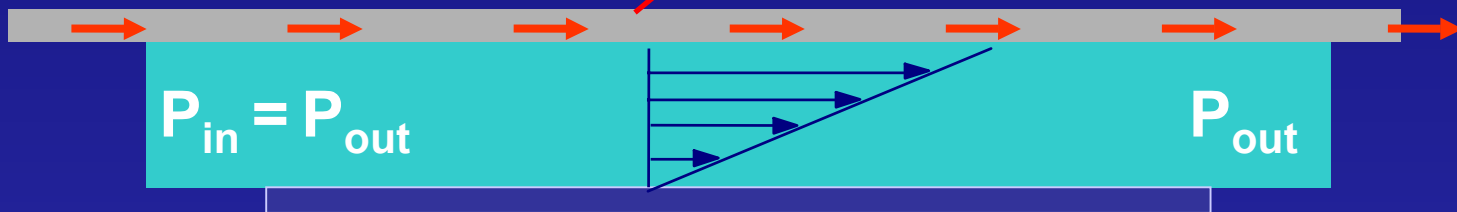
wall+packing = flow resistance



Flow Driving Force at Channel Inlet

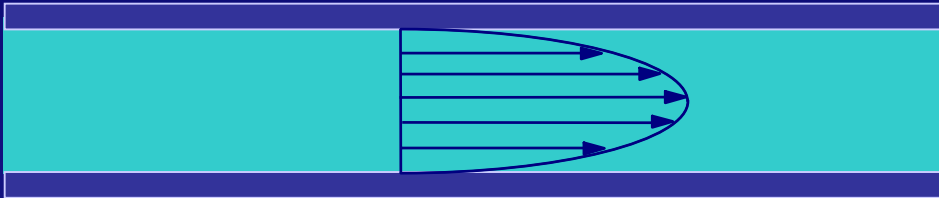
Shear-Driven:

wall="flow driver"



Flow Driving Force along Mantle Surface

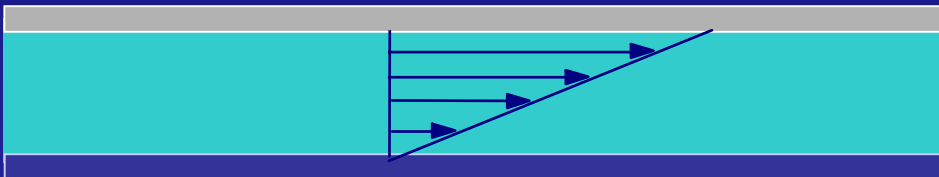
Pressure-Driven:



reduction of d restricted by ΔP -limitation

$$u \sim \frac{\Delta P \cdot d^2}{L}$$

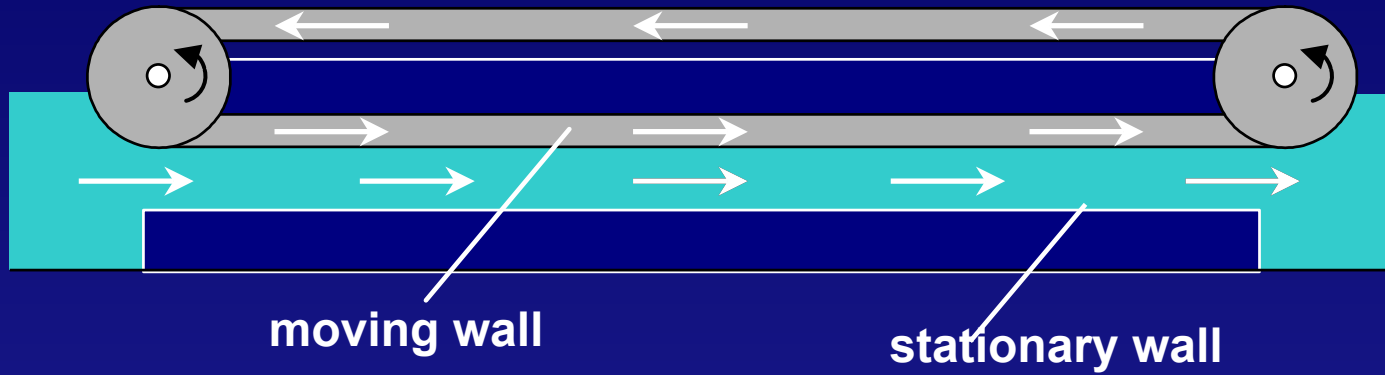
Shear-Driven:



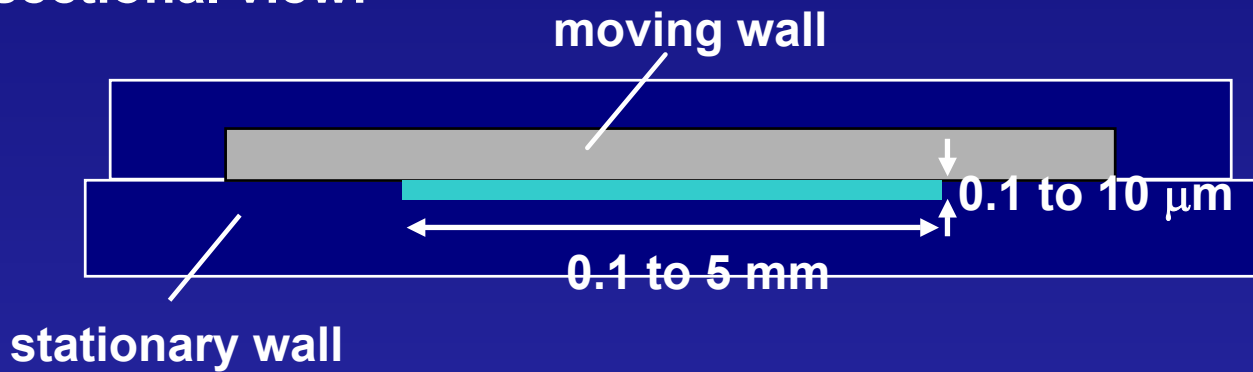
u and d can be selected independently

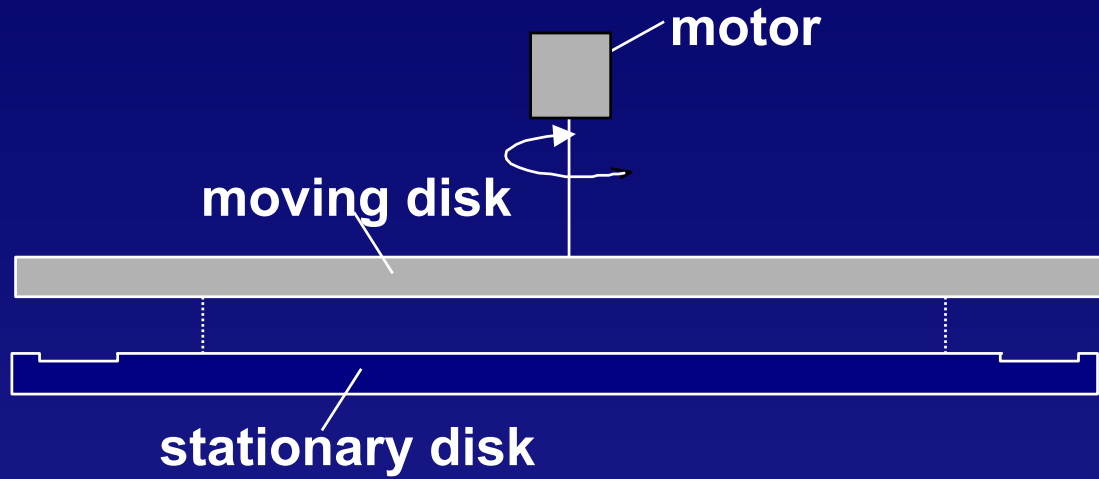
$$u_{\text{avg}} = \frac{1}{2} \cdot u_{\text{moving wall}}$$

Longitudinal view:

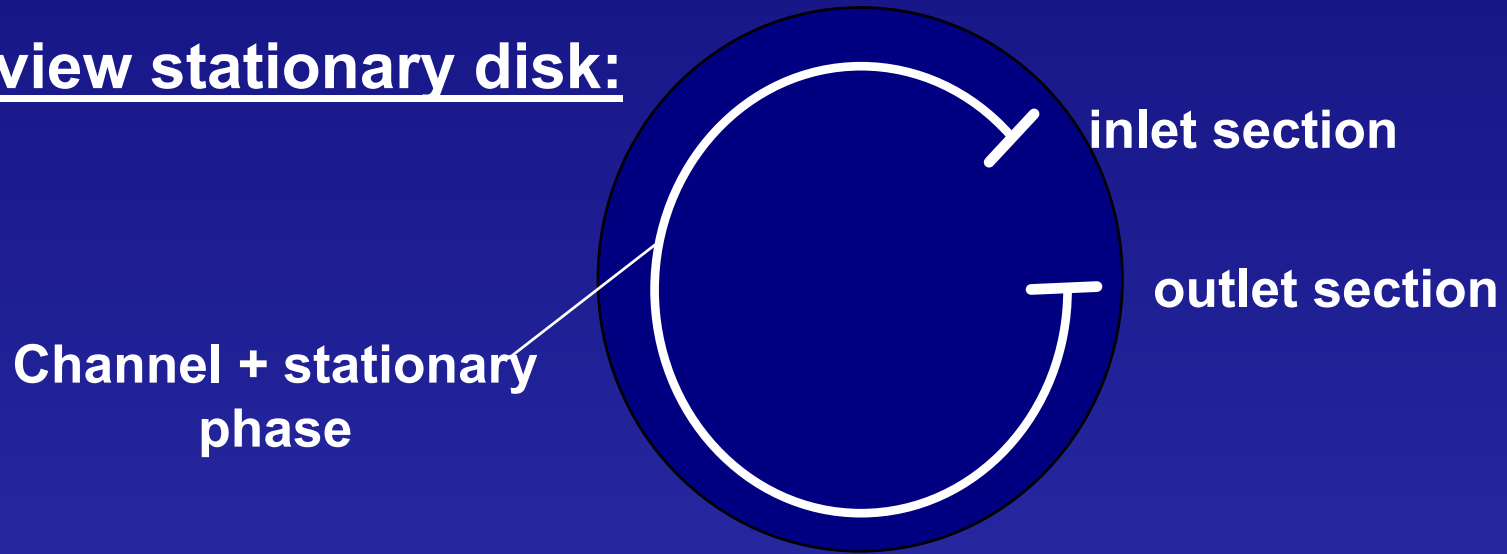


Cross-sectional view:





top view stationary disk:

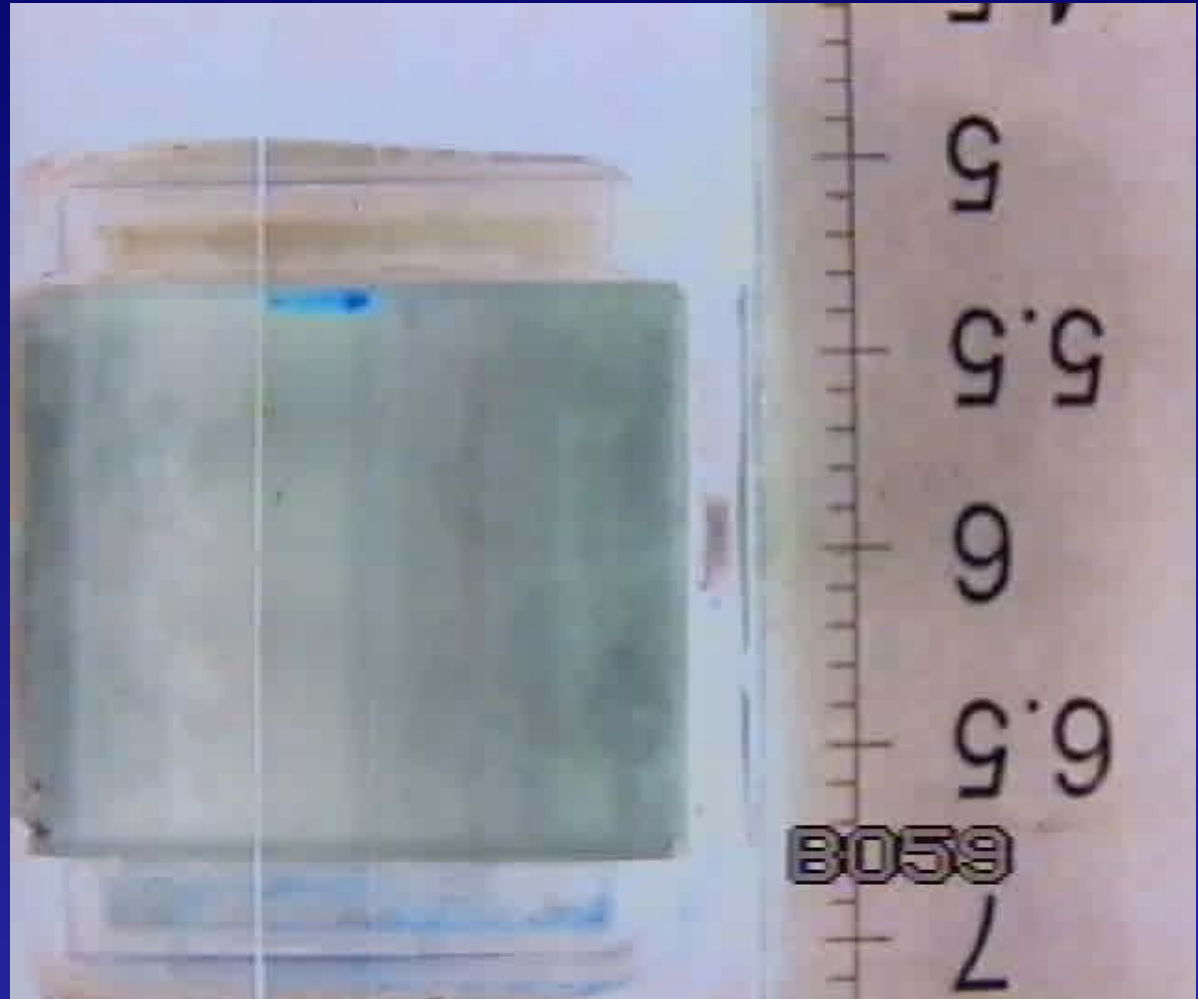


$L = 15 \text{ mm}$

$w = 4 \text{ mm}$

$d = 400 \text{ nm}$

$u_{\text{mean}} = 3 \text{ mm/s}$

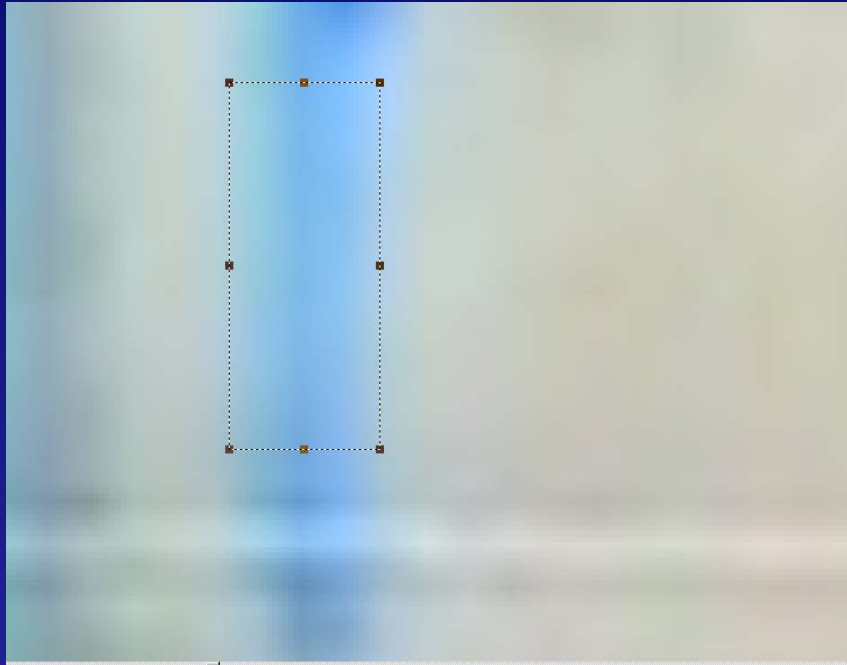


$d = 400$ nanometer, $u_{\text{wall}} = 6$ mm/s

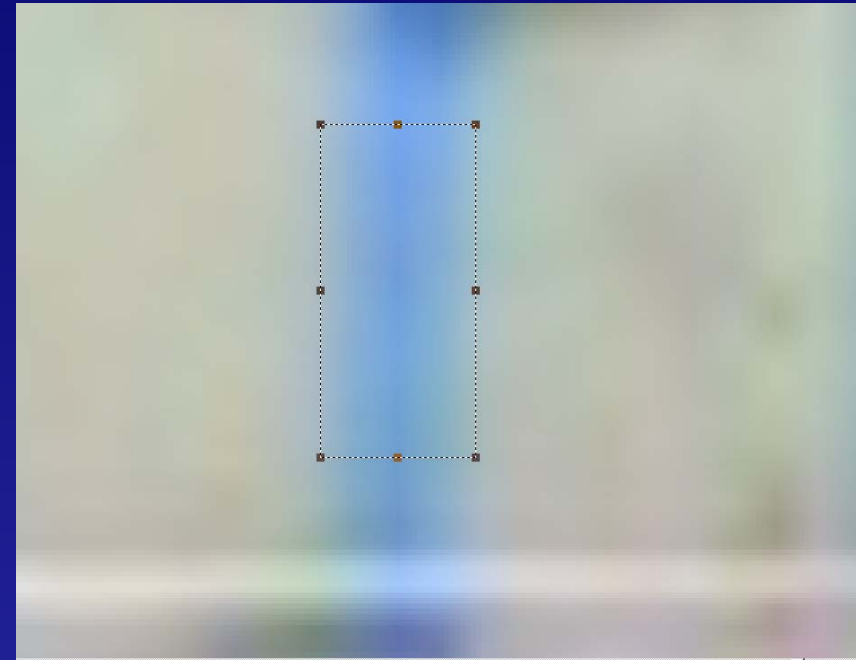


$$u_{\text{avg}} = \frac{1}{2} \cdot u_{\text{moving wall}}$$

$d = 100$ nanometer, $u_{\text{mean}} = 1.4$ cm/s

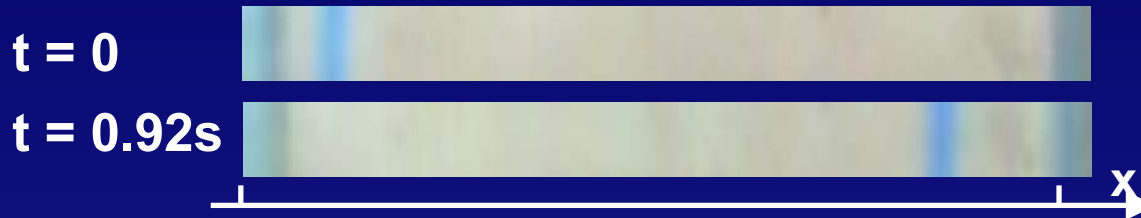


$t = 0$

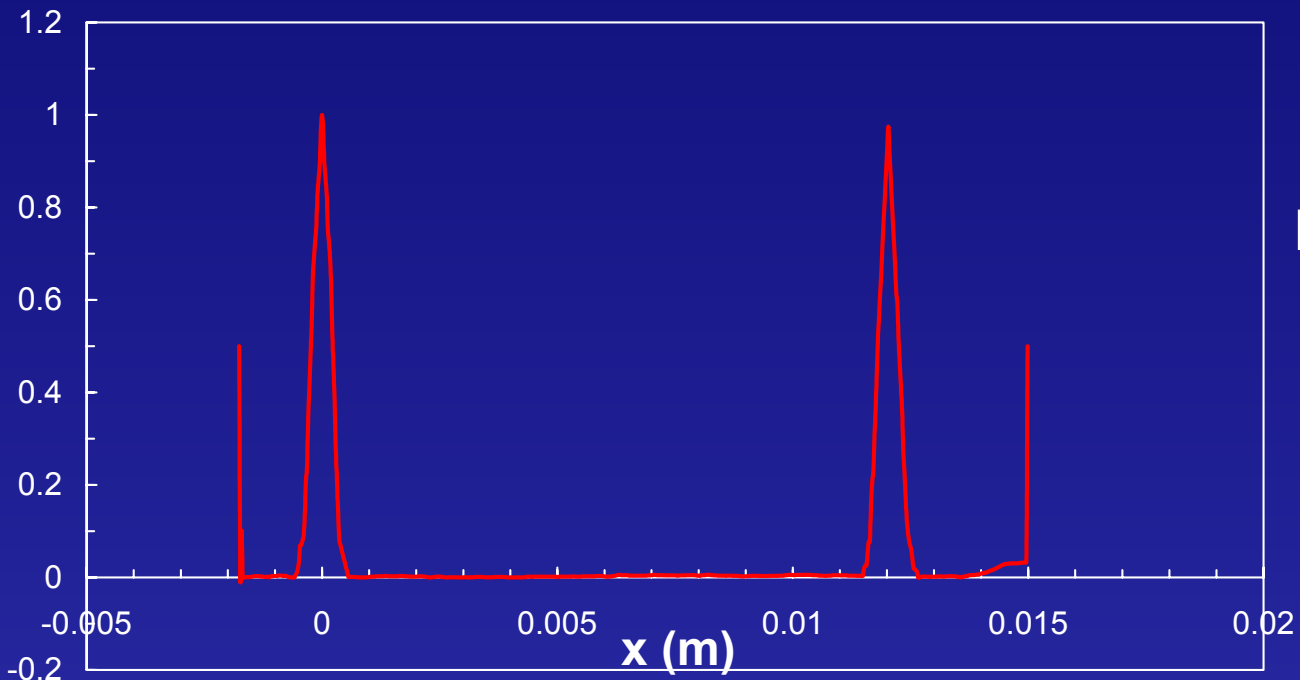


zoom = 800%

$t = 0.92\text{s}$



Plot of Colour Intensity Foto 1 + Foto 2 :

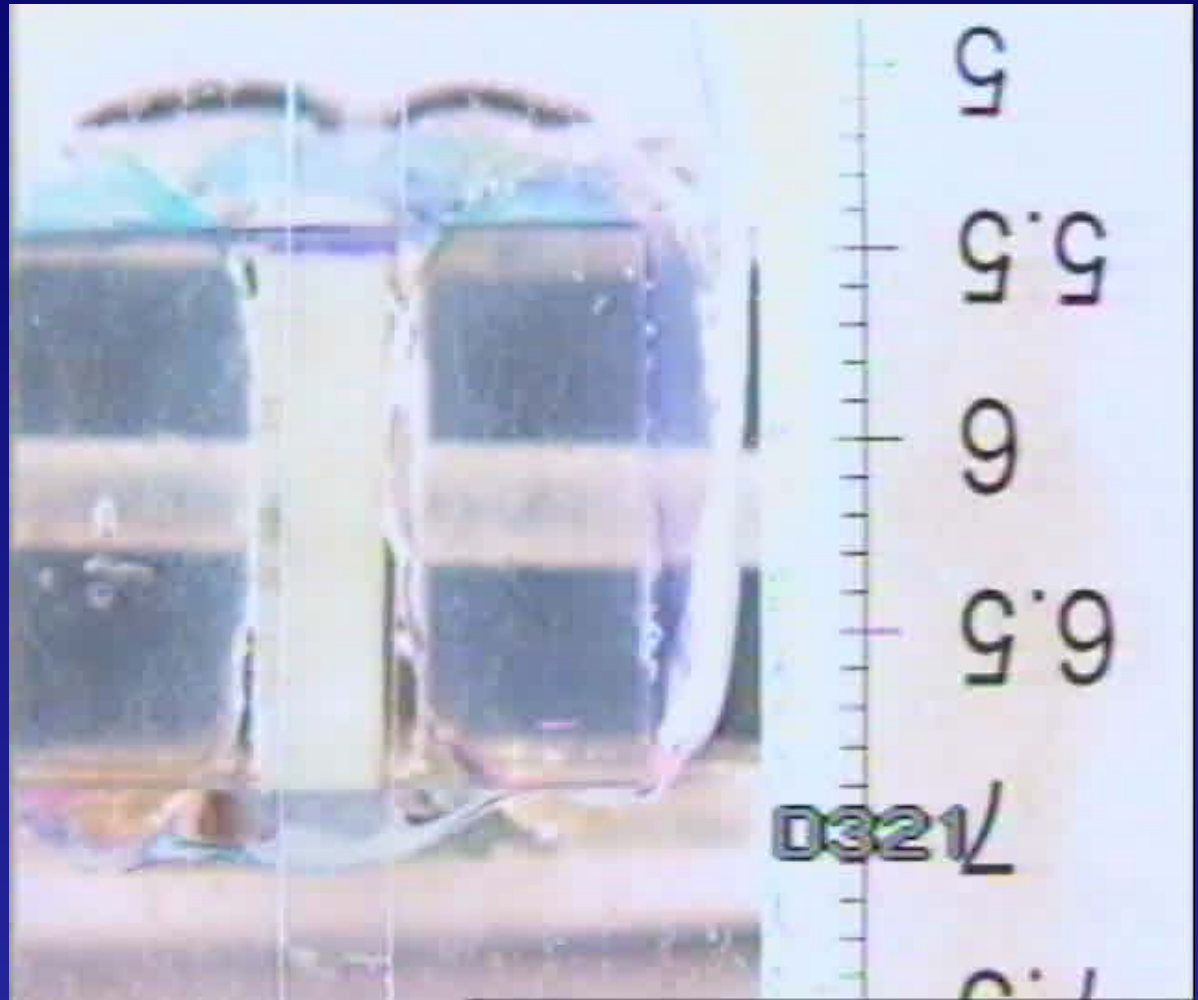


$$\text{HETP} = \frac{\sigma_{\text{out}}^2 - \sigma_{\text{in}}^2}{L}$$

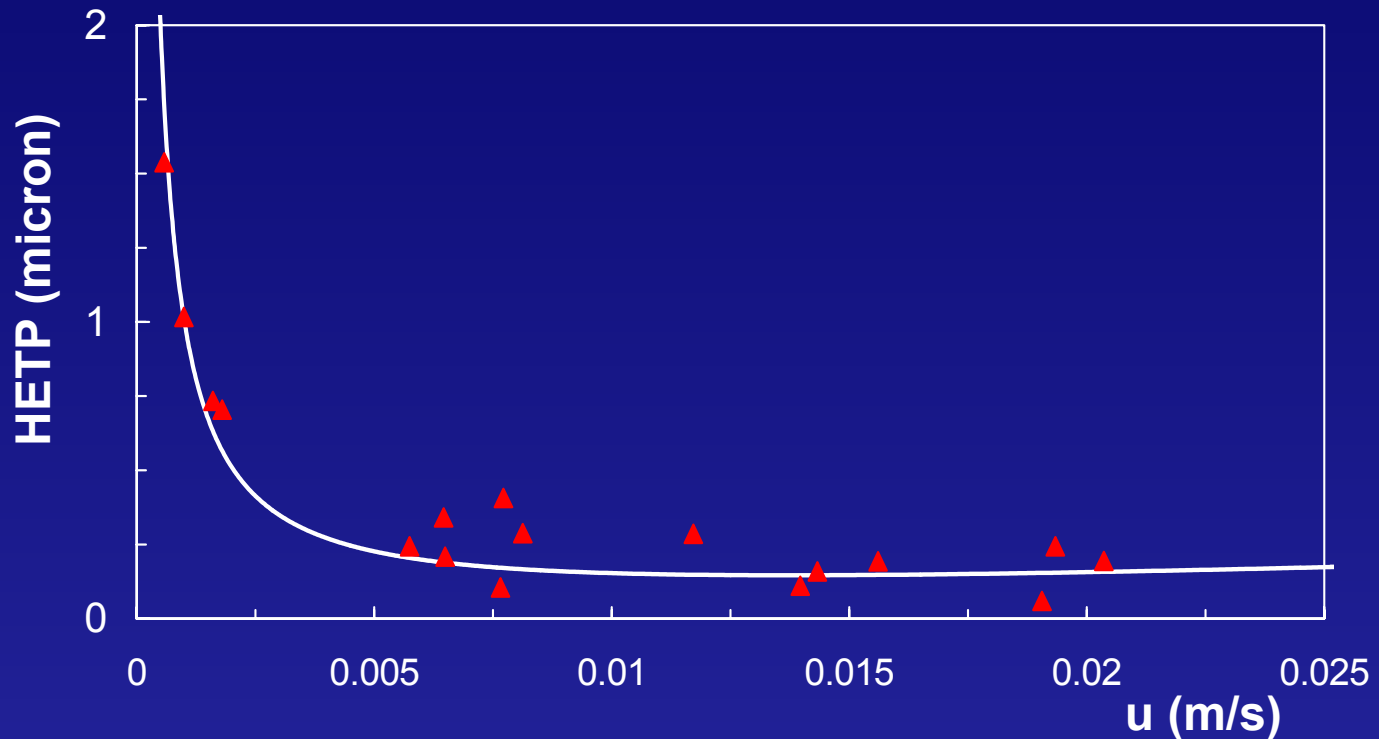
$$= 0.13 \mu\text{m}$$

$d = 100 \text{ nm}$,
 $u_{\text{mean}} = 1.4 \text{ cm/s}$

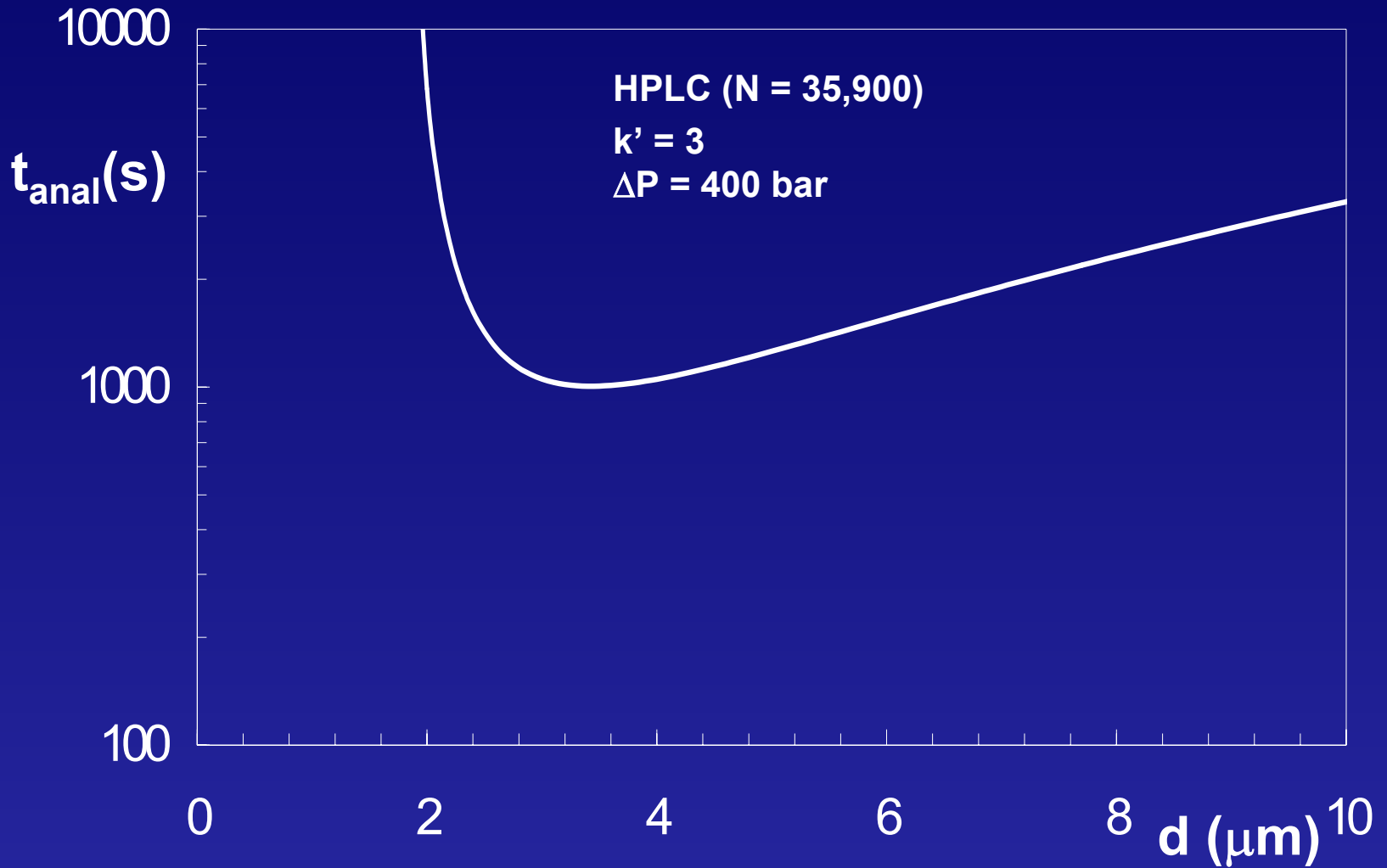
$\Delta P = 3400 \text{ bar !!}$

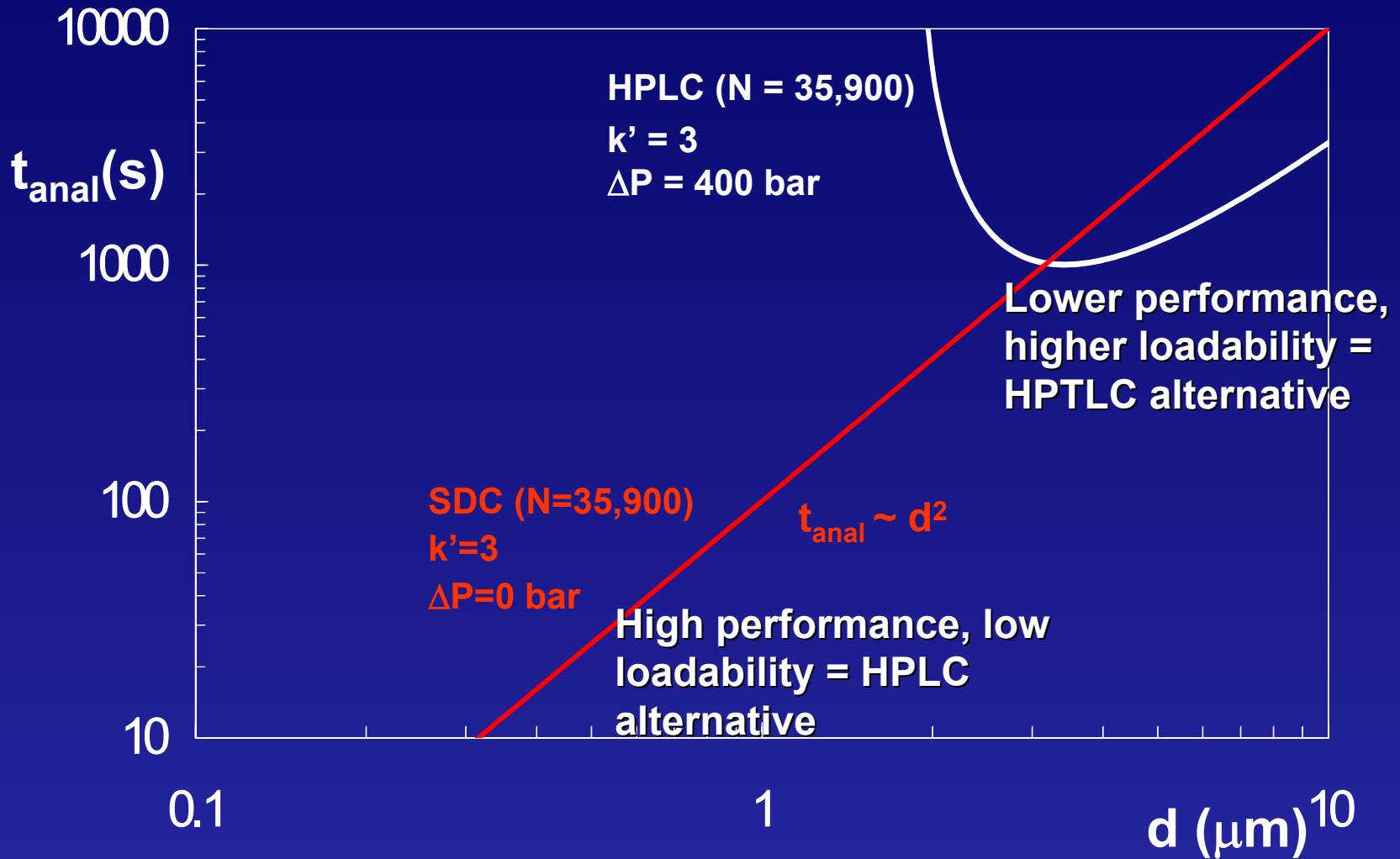


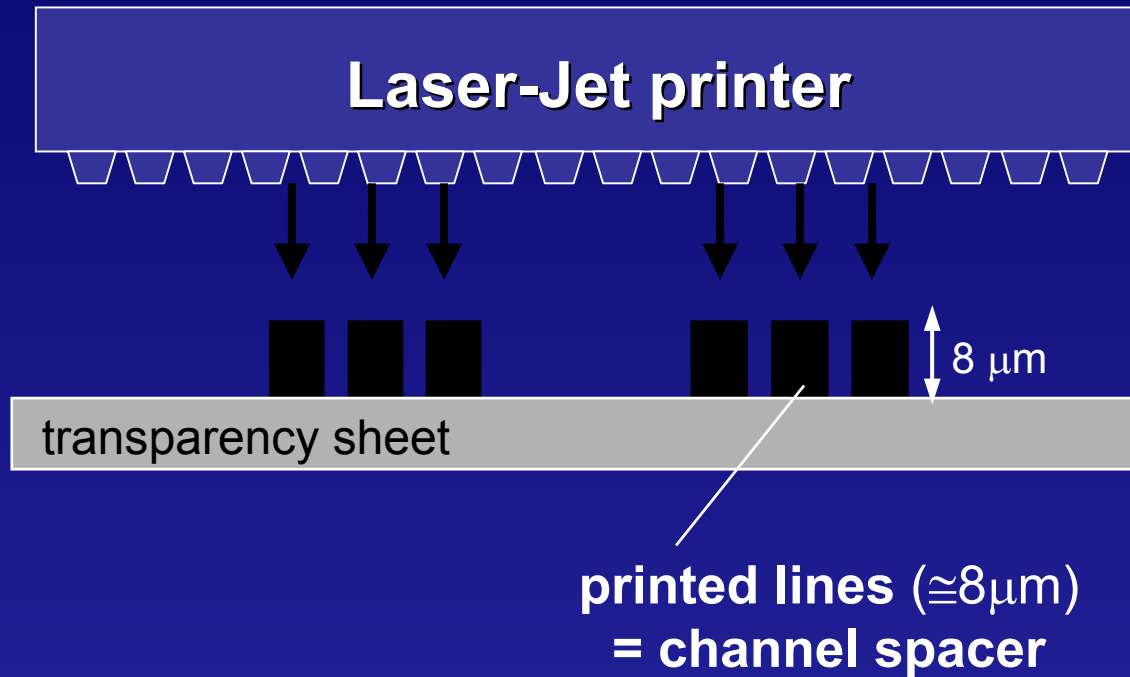
$d = 100$ nanometer, $D_m = 6 \cdot 10^{-10}$ m²/s



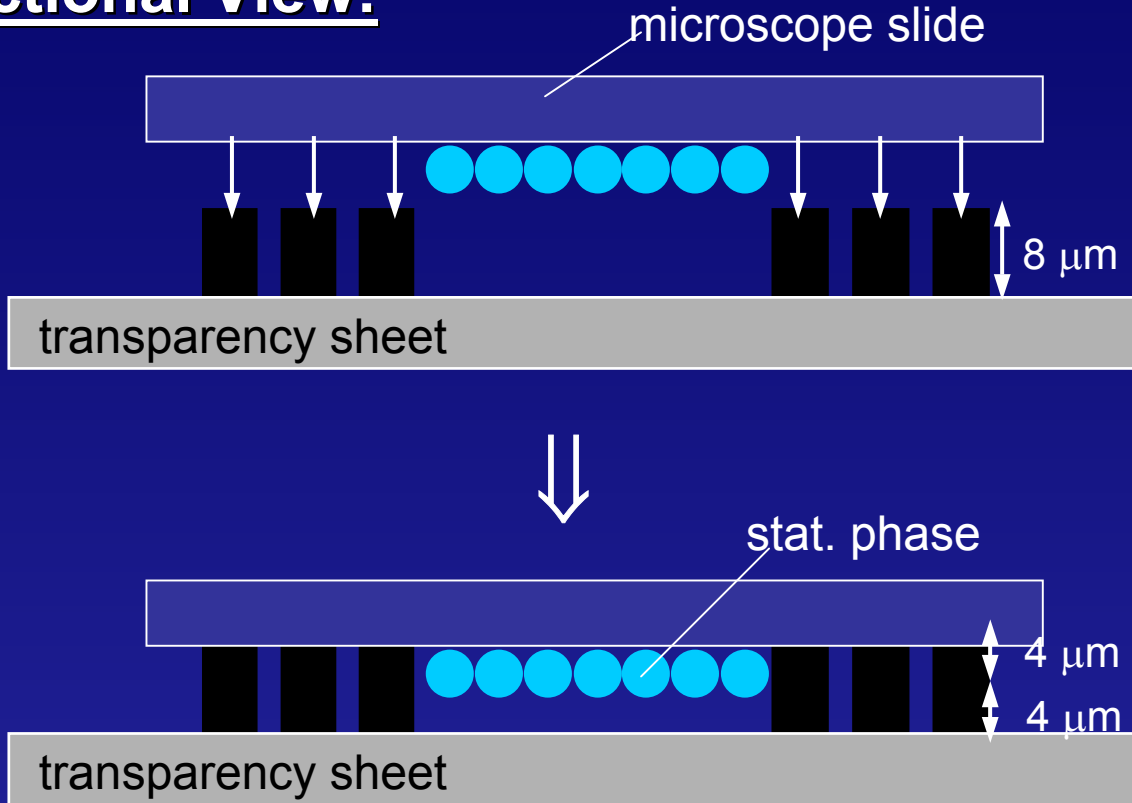
Theoretical:
$$\text{HETP} = \frac{2}{u} \cdot D_m + \frac{2}{30} \cdot \frac{u \cdot d^2}{D_m}$$





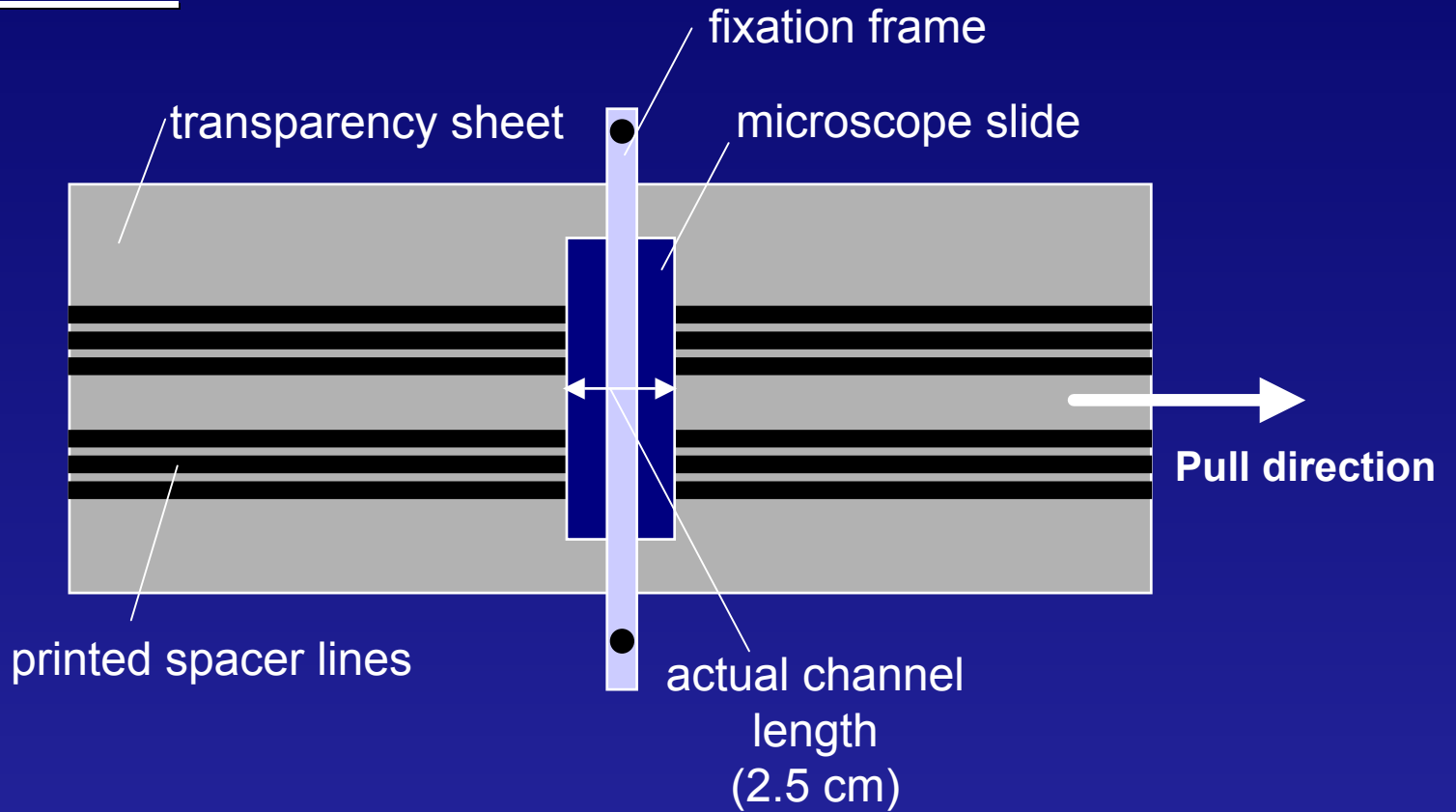


Cross-sectional View:

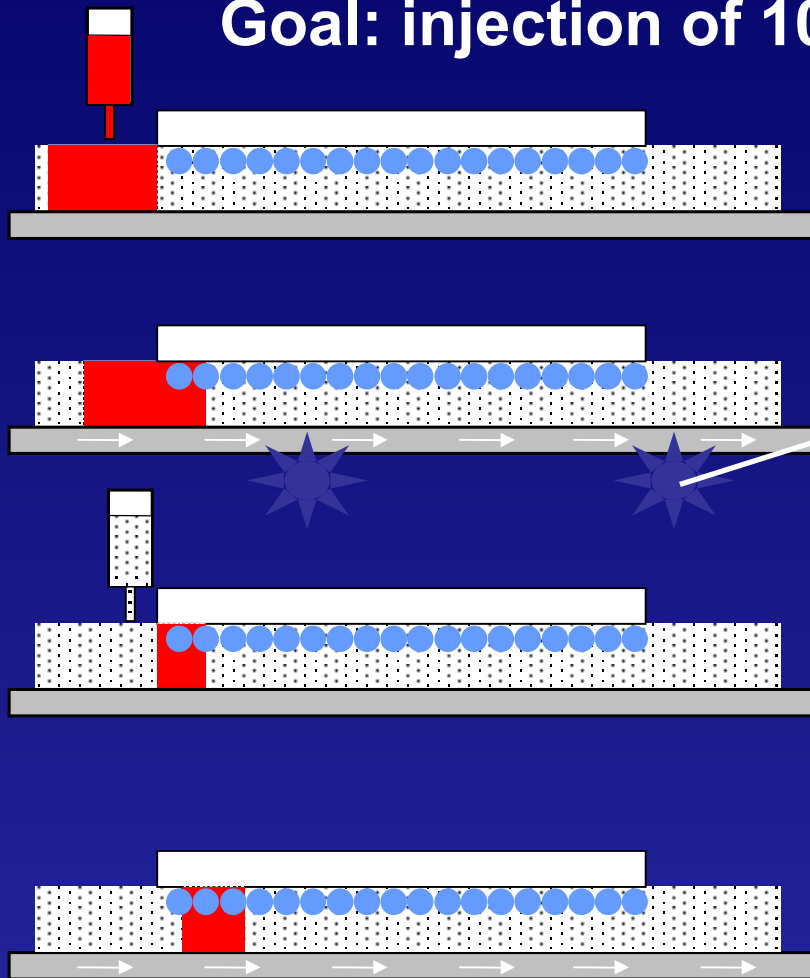


Unsealed channel parts = no problem

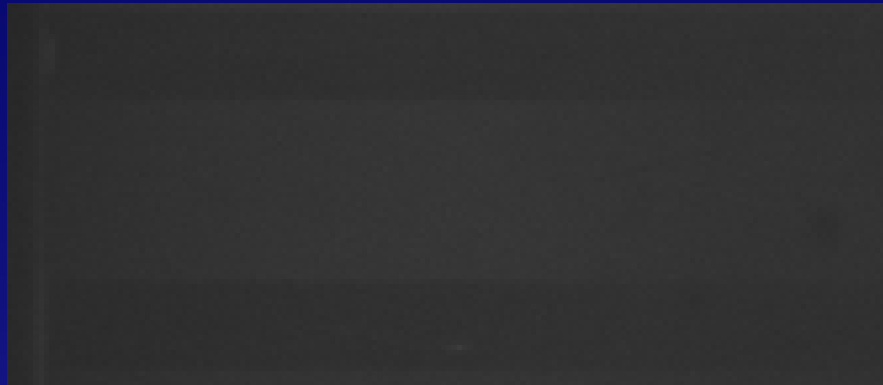
Top View:



Goal: injection of 100 μm bands within 0.5 s



Method: micro-positioner and automatic displacement system (accuracy of 0.5 micron = >99% accuracy)

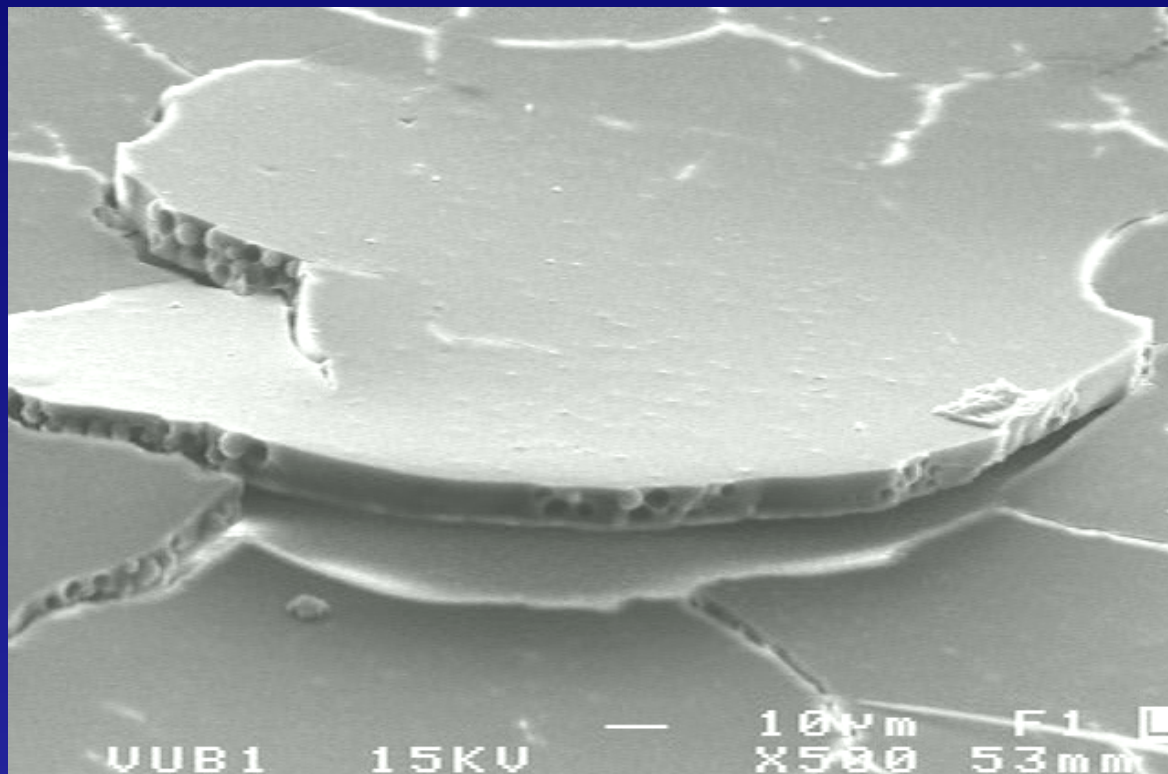


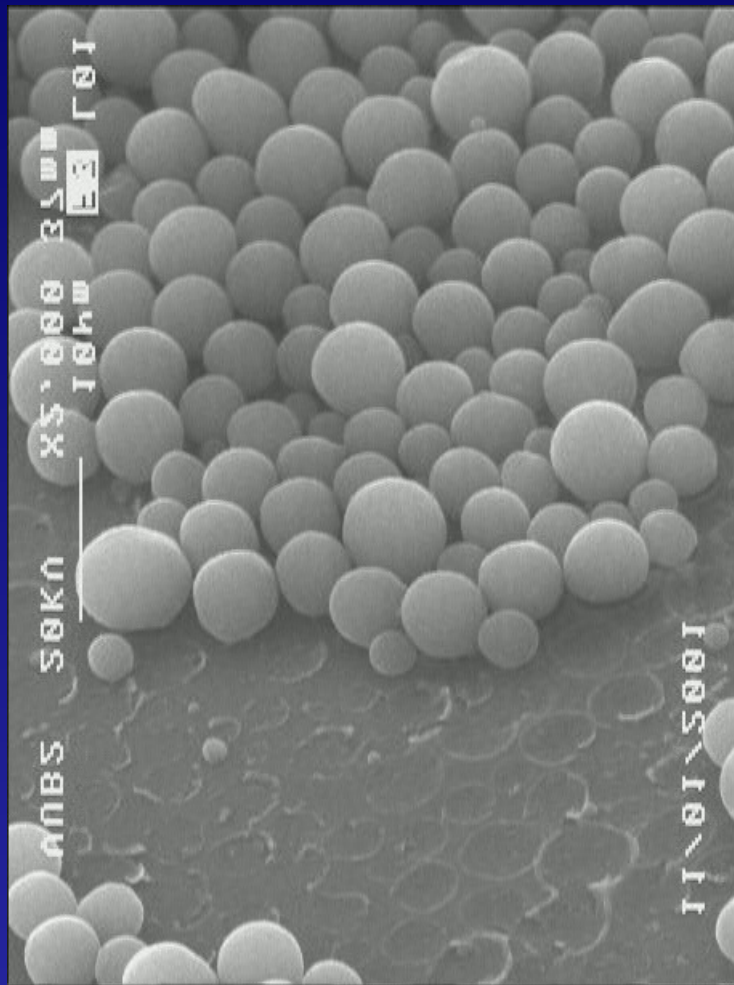
2 mm 



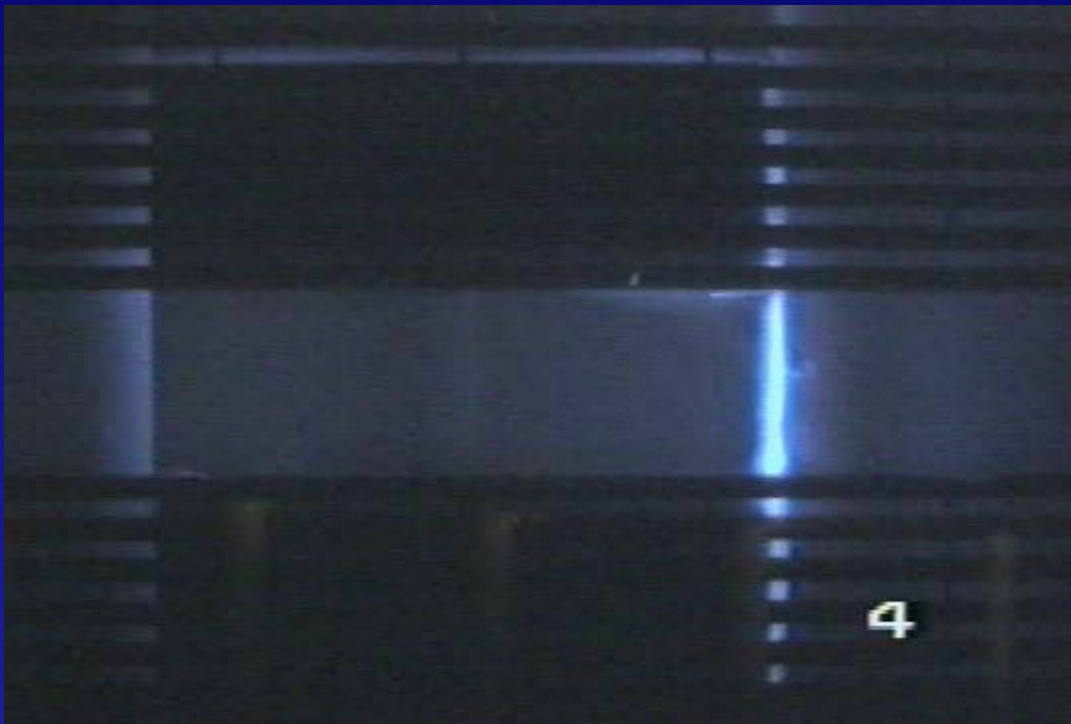
**Channel depth
= 200nm**

Preparation of layer: 4 μm RP-HPLC beads contained within a poly-acrylic polymer matrix



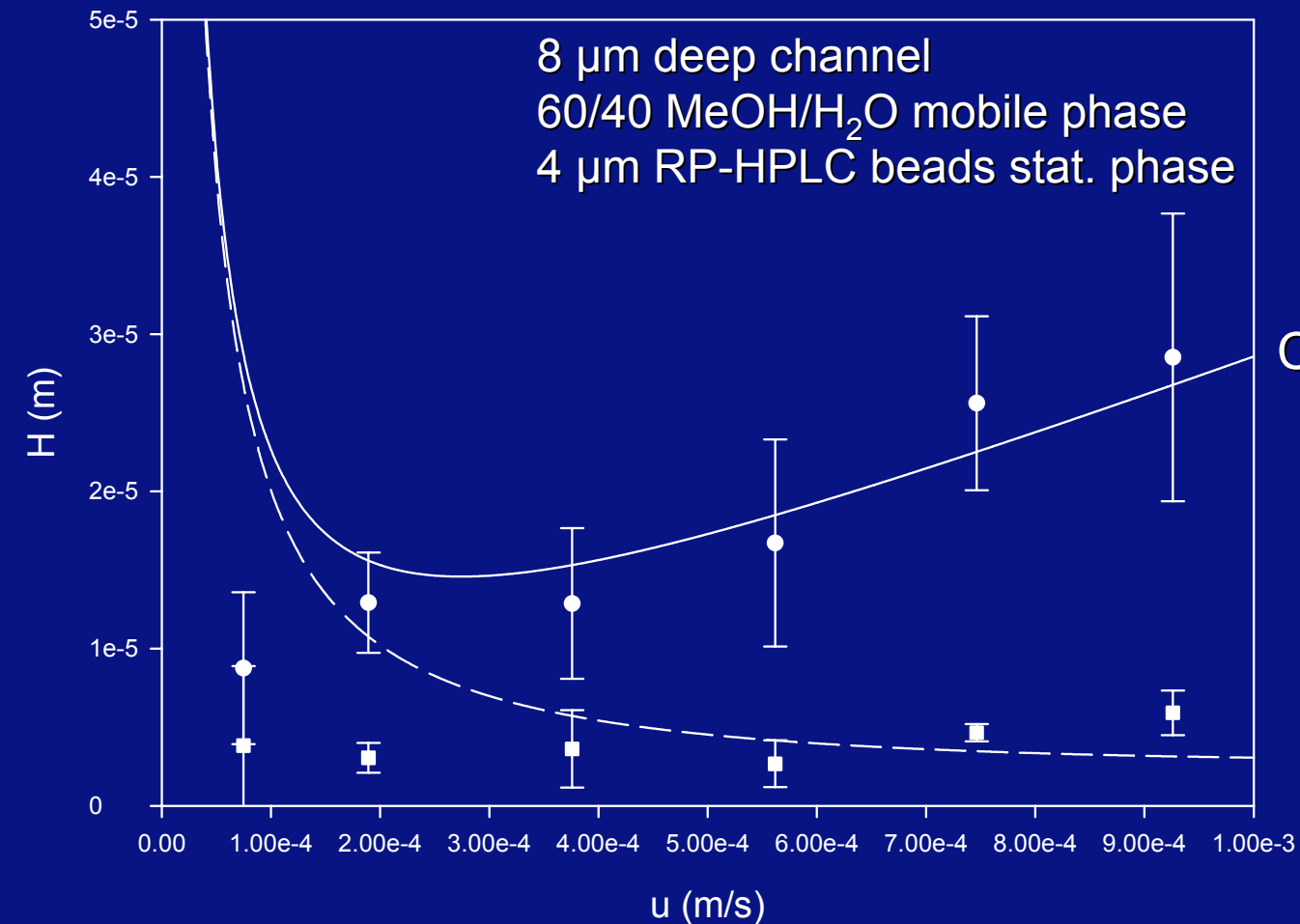


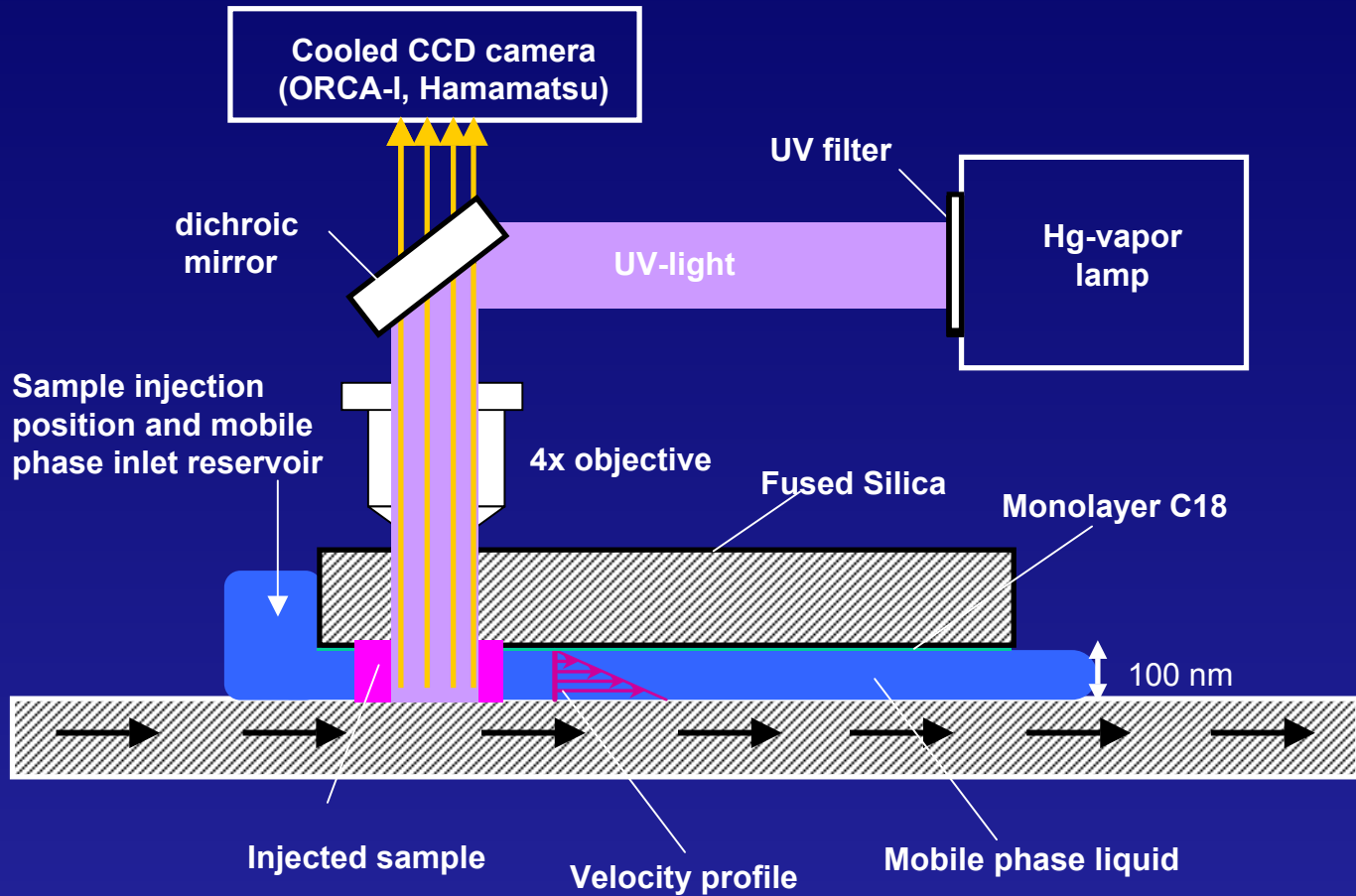
**Monolayer of commercial 4 μm
HPLC-beads immobilised with
Polyethoxysilane layer**



**Mobile phase =
60%/40% water-
methanol**

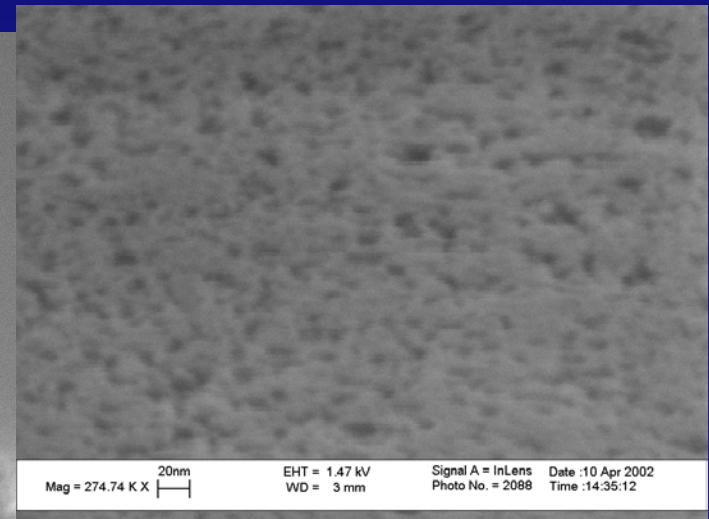
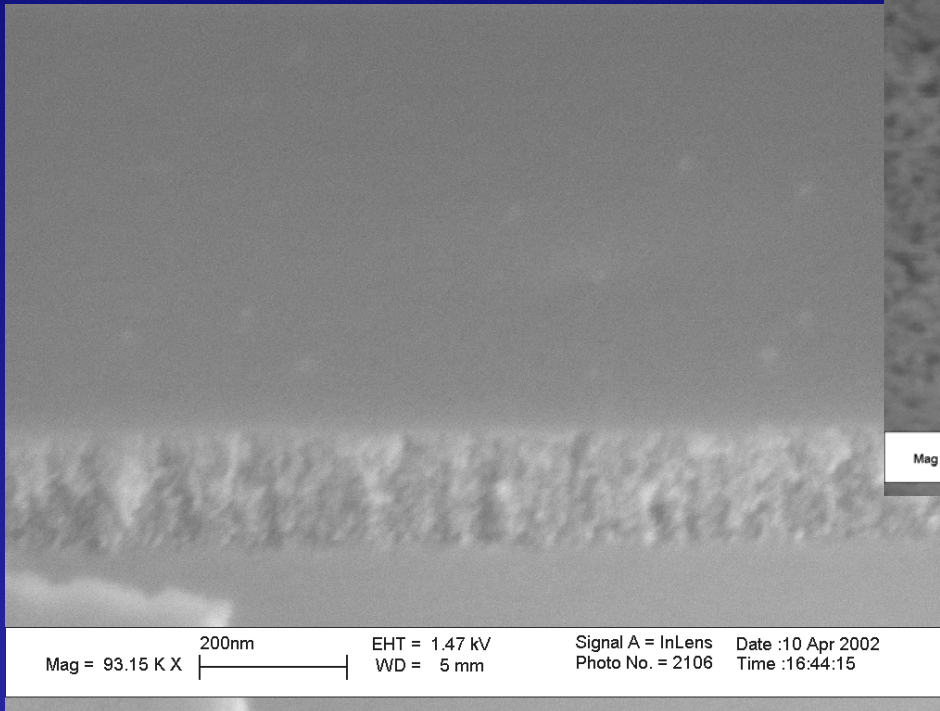
$u = 1.25$ mm/s

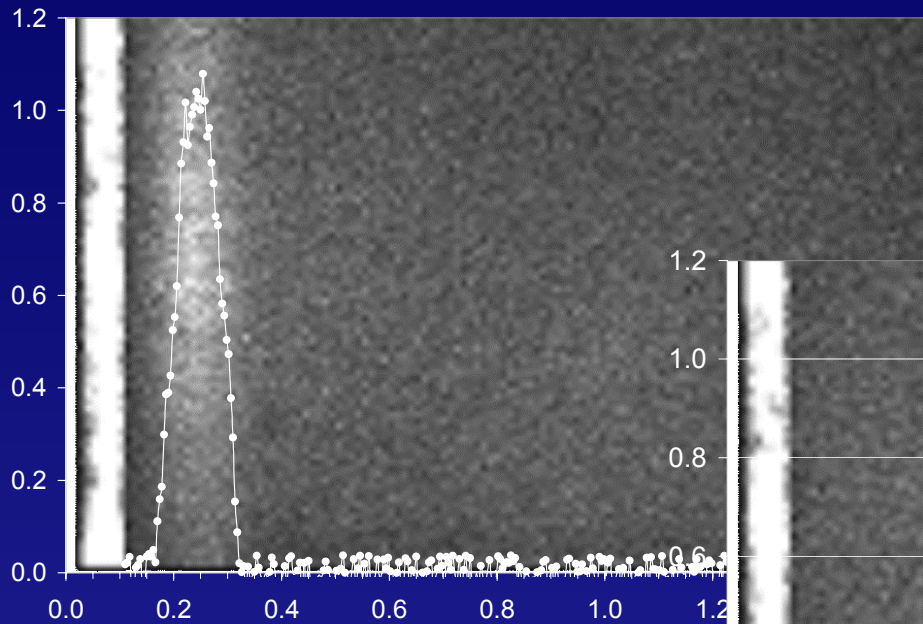




- Electrochemical etching of Si wafer
- Oxidizing Si \rightarrow SiO₂
- Surface modification with C8/C18

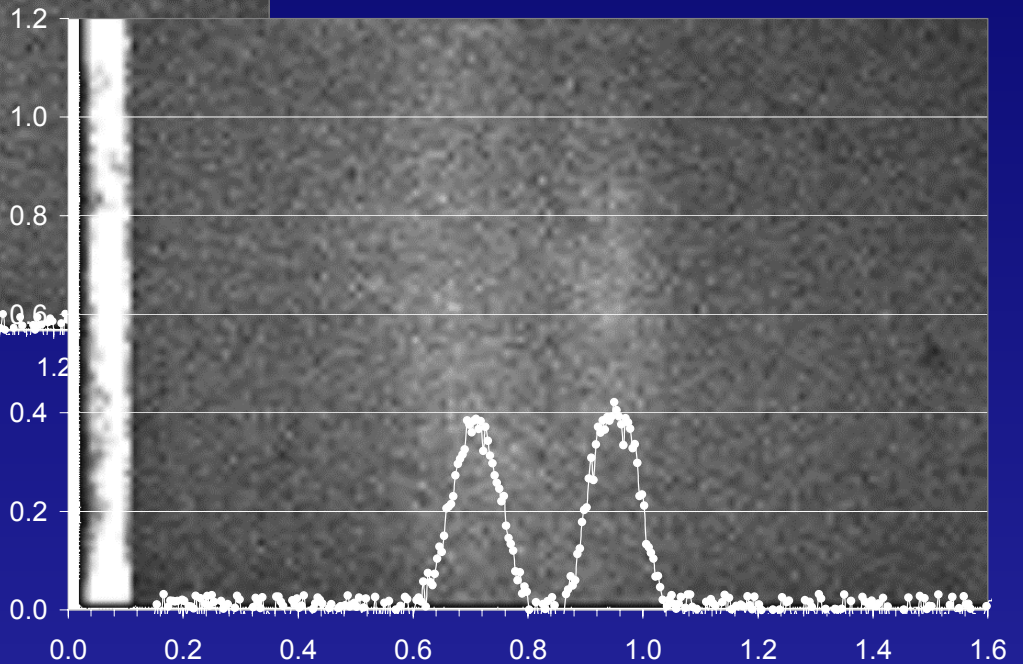
\Rightarrow 100-200 nm thin Porous Silica layer





t = 0 sec

Stationary phase = porous silicon



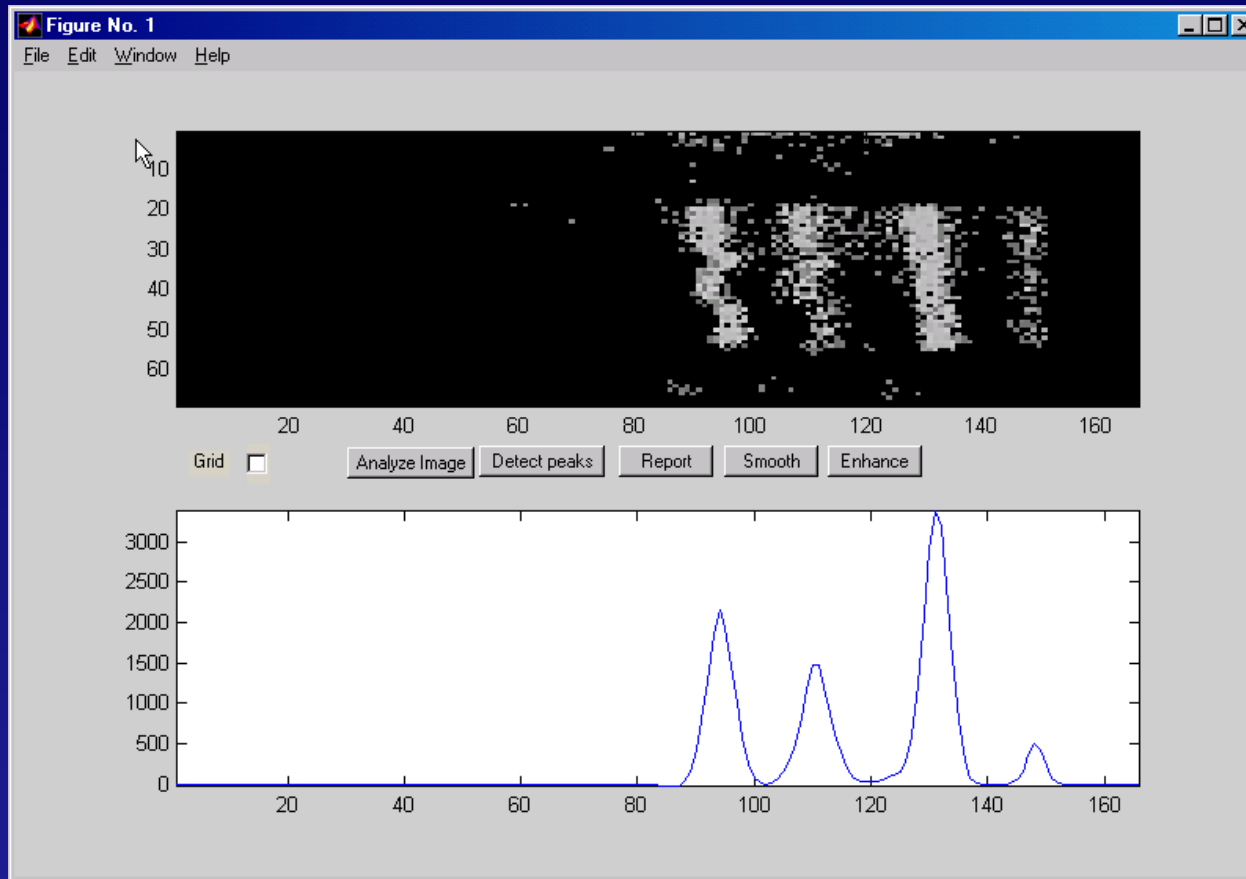
t = 0.05 sec

Separation of four Coumarin dyes on monolayer C8



2 mm 

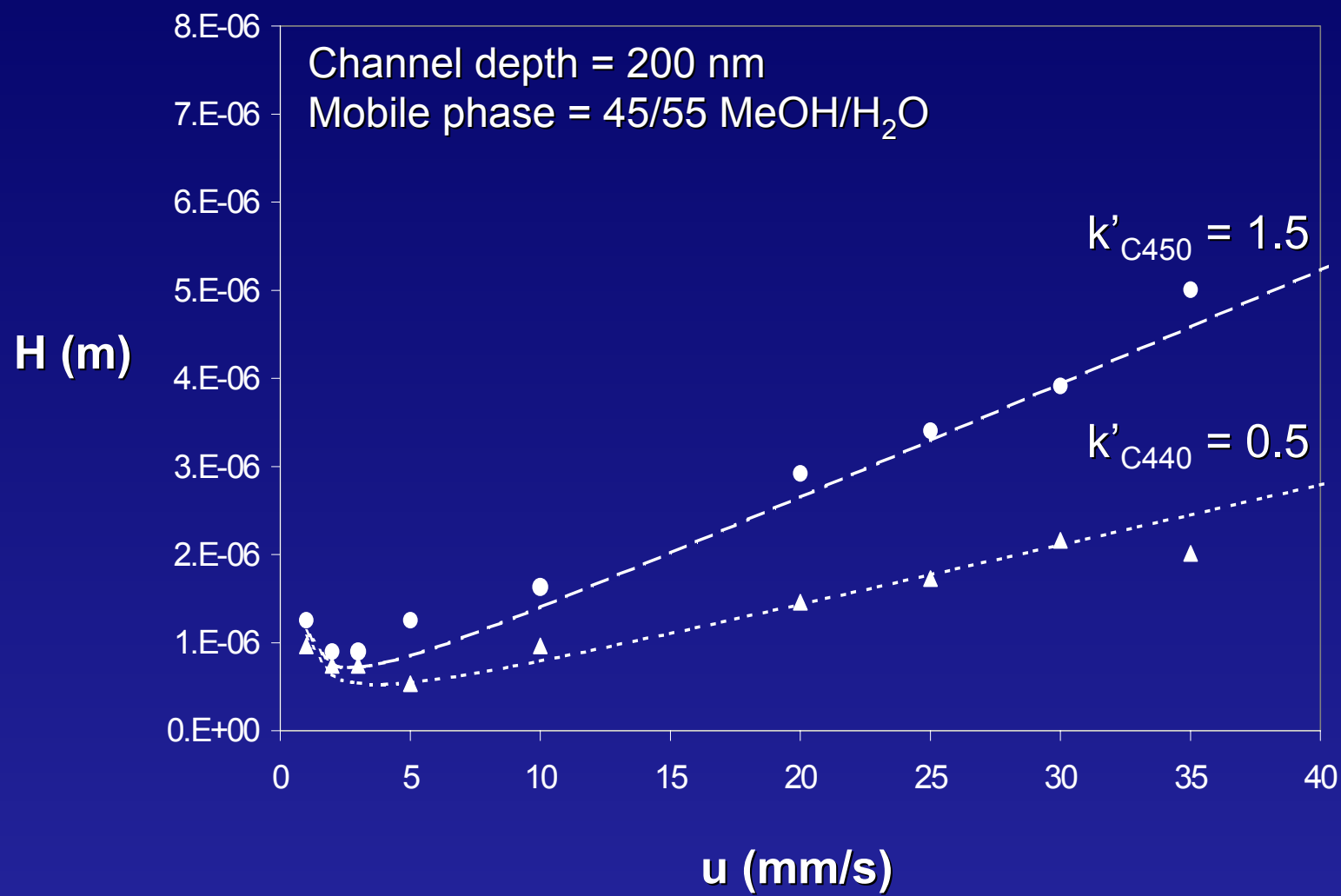
- $u = 2 \text{ mm/s}$
- Mobile Phase = 45/55 % MeOH/H₂O
- $t = 3 \text{ seconds}$



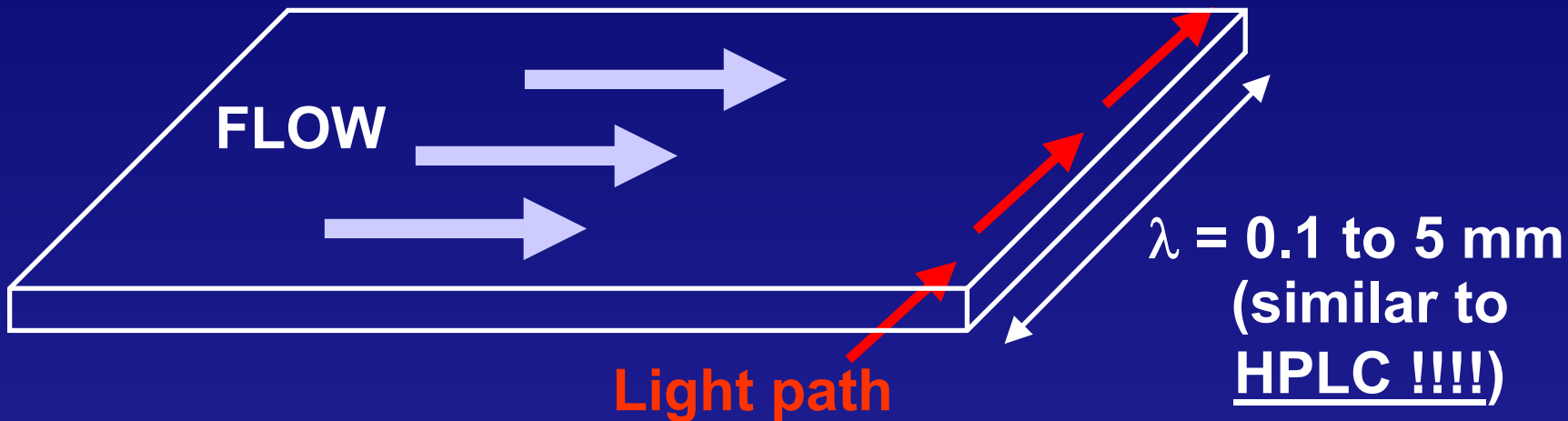
Baseline resolution of 4 compounds in 3 seconds

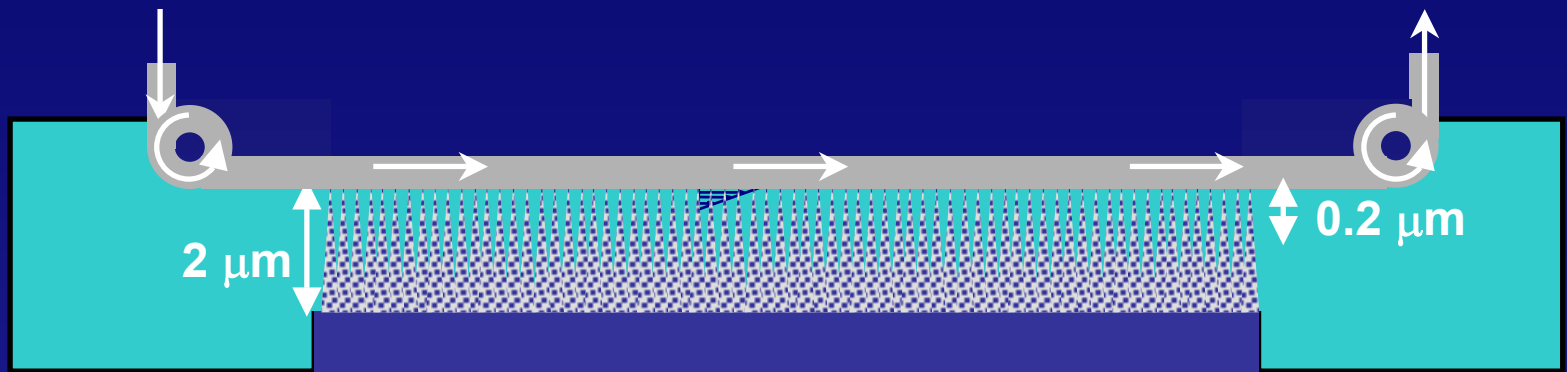
$H \approx 0.5 \mu\text{m} \longrightarrow \sim 2,000,000 \text{ plates/m}$

Channel depth = 200 nm
 Mobile phase = 45/55 MeOH/H₂O



Take advantage of rectangular channel format to benefit maximally from Beer-Lamberts' law: $\text{Log } I_0/I \sim C_{\text{max}} \cdot \lambda$





e.g. use of HPTLC plate only few microns thick which can be processed in a conventional way after separation.

- **Continuous, chromatography enabling flows in channels as thin as 100 nanometer are possible**
- **Perfect predictability of flow rate**
- **Flow system without ΔP or ΔE -limitation**
- **Large potential gain in analysis time or resolution**
- **Offers larger separation window than TLC (all compounds can be eluted at the end of the column)**

The End